

1 **Analysis of the thermal efficiency of a compound**
2 **parabolic Integrated Collector Storage solar water**
3 **heater in Kerman, Iran**

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23 **Abstract**

24 This paper presents an experimental study involving design, manufacturing and testing of a
25 prototype integrated collector storage (ICS) solar water heater (SWH) in combination with a
26 compound parabolic concentrator (CPC). The thermal efficiency of the developed system is
27 evaluated in Kerman (latitude 30.2907°N, longitude 57.0679°E), Iran. A 6-month experimental
28 study was undertaken to investigate the performance of the ICS SWH system. The mean daily
29 efficiency and overnight thermal loss coefficient of each experiment were analyzed to examine the
30 appropriateness of these collectors for regions in Kerman. **The results showed that mirror has the
31 highest mean daily efficiency (66.7%), followed by steel sheet (47.6%) and aluminium foil
32 (43.7%). The analysis of hourly and monthly operation diagrams for variations of water
33 temperature for the developed system showed that by increasing the amount of radiation entering
34 the water heater, the system's thermal efficiency decreases, such that the highest efficiency (during
35 the six-month test period) was in April (66%) and the lowest in July (50%).** The study shows how
36 the temperature gradient between the ambient air and internal water in the storage tank can
37 influence the performance of such systems, and how a controlled amount of hot water withdrawal
38 can affect the system's efficiency.

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43 **Introduction**

44 Industrialization and population growth have caused a dramatic rise in the annual consumption of
45 fossil fuels [1]. Since 1859, most everyday energy usage relied directly or indirectly on fossil fuels.

46 This has caused many environmental problems such as global warming, acid rain, water pollution

47 and increased waste and environmental degradation. By implementing new technologies based on

48 utilizing renewable energy sources, the reliance on fossil fuels dwindles, and the contamination

49 caused by them can also be reduced. Planning and developing new energy sources to replace

50 fossil fuels has increased over the past two decades [2] to be used for different applications [3-6].

51 Solar energy is one of the more reliable sources of renewable energy and can provide 173,000 TW

52 of energy daily to the earth [7]. Although the amount of solar energy that reaches the ground is

53 abundant, its rate per unit area is low. A major challenge is therefore to collect and concentrate

54 this energy efficiently, which is usually done using solar water heaters.

55 Solar water heaters can be divided into two main categories: active and passive systems [8, 9].

56 The active system consists of a solar collector to absorb solar energy and a tank for storing hot

57 water. It includes three main types: direct, indirect, and drain back. In the direct systems, the water

58 circulates directly to the collector. When the temperature of the collector is greater than that of the

59 tank, a pump will circulate the water from the source to the collector. These systems are generally

60 not recommended for climate conditions that lead to cooling of the system or in cases that use

61 heavy or acidic water [10].

62 The indirect systems use antifreeze fluids such as propylene glycol in the collector for the purpose

63 of transferring the heat. The low freezing point of propylene glycol prevents the freezing of the

64 system and allows the solar systems to be used under conditions below 0 °C. The indirect systems

65 prevent the reverse flow of thermosiphon by using a one-way valve at night [11]. The third type

66 of active system is the water-back system. Water-back systems use water for heat transfer. To
67 prevent the risk of frost when the collector temperature is lower than the tank temperature, the
68 pump is switched off and the water inside the system returns to the reserve. The space inside the
69 collector is then filled with air, which protects against the freezing of the system [12].

70 Although active systems are relatively easy to set up, efficient passive systems are also utilized
71 based on their own capabilities. These are divided into two systems: Thermosiphon and ICS. The
72 Thermosiphon system uses high radiation absorption when it is heated (decreasing density). In this
73 system, a storage tank is installed at an altitude higher than the collector. When the water is heated,
74 it becomes lighter and naturally flows to the highest point inside the supply. The cold water from
75 the bottom of the source flows through the pipes to the bottom of the collector and creates a natural
76 circulation in the system. Circulation in the system stops when the temperature inside the collector
77 becomes lower than the temperature inside the tank. This prevents heat transfer from the system
78 to the environment at night when the temperature of the collector is lower than that of the supply
79 [13]. In the ICS type of passive SWHs, the storage tank and the collector are not separated from
80 each other. The cold water is directly connected to the collector and is heated by the sun. Unlike
81 other systems, hot water remains in the collector until it is consumed and then directly used by the
82 collector. ICS systems require larger storage sources (to increase radiation absorption capacity)
83 than conventional systems, which also protect the system against frost [14].

84 Many amendments have been made in recent years to various parts of the ICS system so that the
85 maximum absorption of radiation energy and the lowest thermal loss can be achieved
86 simultaneously for the system. From the early 1800s, a number of solar concentrators were built
87 to achieve higher temperatures and steam production [15]. The first ICS SWH was developed in
88 the southwest of the USA at the end of the 18th century. In 1982, Tiller and Wochatz analyzed the

89 performance of ICS water heaters in warm climates and discovered that ICS' performance is higher
90 than flat plate SWHs [16]. In another experimental study, a solar water heater (SWH) with a plastic
91 water bag (PWB) was designed and developed that produced desirable results in different weather
92 conditions [17]. In 1984, Faiman provided a standard method for calculating the efficiency of ICS
93 SWH water heaters [18]. This methodology helped Waller et al. [19] to analyze the rate of heat
94 loss in ICS solar water heaters. Rommel [20] concluded that transparent insulating materials can
95 increase the efficiency of ICS water heaters. Mohammed et al. [21] constructed an ICS SWH with
96 a buffer inside a repository, which showed better efficiency in comparison to non-buffer SWHs.
97 Tripanagnostopoulos & Souliotis [22] designed four different models with asymmetric geometries
98 by analyzing the system's heat loss during the night and their thermal performance. The results
99 were compared with the CPC concentrator. It was shown that the SWH with asymmetric geometry
100 had less thermal dissipation at night than the SWH with symmetrical geometry, and the system
101 with symmetric geometry had a higher efficiency than the others [22].

102 Regional studies were also planned to provide a database for governments and decision makers to
103 invest their budget wisely. One of these studies examined the resistance of a water heater to frost
104 in northern European climate conditions [23] to establish an economic evaluation of ICS water
105 heaters in mass production. In this study, the researchers analyzed the return on investment (ROI)
106 and showed that a large number of ICS water heaters was economical and also technically feasible
107 [24]. In 2003, Souliotis & Tripanagnostopoulos [25] conducted experiments on various ICS solar
108 water heaters. They constructed three water heaters with different CPM profiles with different
109 dams and showed that the SWH with a 30-cm droplet dryer had better thermal performance. In
110 another study, the temperature classification of ICS water heaters with a horizontal storage tanks
111 was experimentally evaluated to investigate the effects of multiple combinations of such systems

112 [26]. In 2008, Souliotis and Tripanagnostopoulos [27] analyzed the general distribution of water
113 in the tank wall of the heater. The results illustrated that the upper part of the source reserve
114 absorbed the majority of the reflected radiation. The first study on optical performance of an ICS
115 heater was published in 2015 [28]. The results of this study provided a significant improvement in
116 optical efficiency and the distribution of radiation absorption with various solar slope angles. In
117 the following year, the same researchers optimized the geometrical characteristics of an ICS SWH
118 system in order to find the optimal thermal performances [29]. Most recently, Harmim et al. [30]
119 simulated and tested an ICS SWH for integration into a building façade. Their system had a daily
120 efficiency between 36.4 and 51.6% and its thermal loss coefficient during night-time was between
121 2.17 and 3.12W. The results from these studies showed that different regions receive different
122 amounts of energy based on their assigned latitudes.

123 Because of their location, some countries have benefited from a greater amount of solar intensity
124 [31]. One of these countries is Iran, which is located at a latitude of between 25 to 40 degrees
125 North. The amount of solar radiation in Iran is estimated to be between 1800 and 2200 kWh/m²
126 per year, which is higher than the global average [32]. In Iran, on average more than 280 days per
127 year are sunny, which is higher than a vast majority of the countries within Europe [33]. Therefore,
128 it can rely on different forms of solar energy solution to generate electricity and provide the heating
129 requirements for residential homes. By analyzing all sites of Iran, Shiraz, Yazd and Kerman have
130 areas with higher solar radiation, as illustrated in the solar GIS map [34]. Unfortunately, due to the
131 fact that Iran has one of the largest endowment of oil and natural gas resources in the world [35],
132 most of the population of these regions are already utilizing natural gas for heating water [36]. In
133 this regional study, the Kerman Province, which has 23 cities and encompasses more than 11% of
134 Iran's terrestrial area, was selected for investigation [37]. In addition, its vicinity to the Lut Desert,

135 lower population density and abundance of remote villages compared to its neighboring provinces
136 are other features making it apt for investigating solar water heaters [38]. Although the
137 technologies to generate electricity such as photovoltaic (PV) systems have started to be used in
138 Iran, the applications of solar water heaters has received little to no attention. This technology can
139 be widely used to provide hot water for domestic and industrial consumers in Kerman province,
140 as well as other regions around the world with similar conditions that have considerable solar
141 radiance (see Fig. 1) [39].

142 The system presented in this paper has several advantages: it is economically affordable,
143 structurally robust and easy to manufacture and install using local material, especially in
144 developing countries. Although there have been some studies on ICS SWH systems in some parts
145 of the world, the findings cannot be easily extended to other specific regions. Furthermore, long-
146 term experiments on the performance and efficiency of ICS SWH systems is rare in the literature.
147 In the following sections, the design and testing processes for an ICS SWH system for Kerman
148 Province is presented and discussed.

149

150 **2. Geometric design of the ICS SWH system**

151 **2.1. Storage tank**

152 The storage tank is the central part of the ICS SWH system. Its function is to absorb solar radiation
153 and transfer the thermal energy to the stored water. The size and shape of a storage tank have an
154 important effect on the absorption of solar energy. The greater the area of the storage tank exposed
155 to the sun, the less time it takes to warm the water. However, in a normal climate, a high-surface
156 storage tank will lose a significant part of the energy through heat transfer and radiation with long
157 wavelengths, and mostly during the night due to heat loss to the surroundings [40]. According to

158 the study carried out by Keshavarzia et al. [41], water consumption in rural regions of Iran is about
159 120 l/d per person. In order to supply this amount of household water, the diameter and length of
160 the storage tank were chosen as 30 cm and 200 cm respectively, resulting in a storage tank volume
161 of just over 141 l. The storage tank should be made of high conductivity material such as
162 aluminum, copper or galvanized iron/steel [42]. The thickness of the sheet should be chosen such
163 that it withstands the pressure of the water, in addition to having low thermal resistance against
164 heat transfer [40]. Considering the available sheets in the market, a thickness of 1 mm was chosen.

165

166

167 **2.2. Geometry of CPC concentrator**

168 A symmetric CPC concentrator was selected to achieve the highest efficiency based on literature
169 [22] (see Fig. 2).

170

171 Fig. 2 - Cross-section view of a symmetric CPC [22]

172

173 In such models, the concentrator is composed of four parts: AB, BC, C'D, and DA' (see Fig. 2).

174 The parabolic equations for the different parts are as follows [22]:

175 Part 1 (AB):

$$176 X = -r_T[1 + \pi \sin \psi / (1 + \cos \psi)] \quad (1)$$

$$177 Y = -r_T[\pi \cos \psi / (1 + \cos \psi)] \quad (2)$$

178 Part 2 (BC)

179 $X = -r_T(\sin \omega - \cos \omega)$ (3)

180 $Y = -r_T(\cos \omega + \sin \omega)$ (4)

181 Part 3: (C'D):

182 $X = r_T(\cos \acute{\omega} + \acute{\omega} \sin \acute{\omega})$ (5)

183 $Y = r_T(\sin \acute{\omega} - \acute{\omega} \cos \acute{\omega})$ (6)

184 Part 4 (DA '):

185 $X = r_T \pi \cos \acute{\psi} / (1 + \cos \acute{\psi})$ (7)

186 $Y = r_T [1 + \pi \sin \acute{\psi} / (1 + \cos \acute{\psi})]$ (8)

187

188 where: ω : involute concentrator's angle; $\acute{\omega}$: upper involute concentrator's angle; ψ : parabolic
 189 concentrator's angle; $\acute{\psi}$: upper parabolic concentrator's angle; $\acute{\psi} = \psi = 58$ degrees and $\omega = \acute{\omega} =$
 190 90 degrees [22].

191 The width of the concentrator ($W\alpha$) is $AA' = BD = 2\pi r_T$ [22]. As the radius of the storage tank
 192 (r_T) was chosen as 15 cm, $W\alpha$ is 95 cm and the length of the aperture (L_α), which is equal to the
 193 length of the storage tank (L_T), is 200 cm. This results in the aperture area (A_α) of 16,500 cm² for
 194 the SWH system. To obtain the exact dimensions of the curves, a computer program was written
 195 based on the above equations to calculate the coordinates of the points for each degree of variation
 196 of ψ and ω . The system specification is presented in Table 1.

197

198 Table 1: Configuration details of the tested ICS SWH

199

200 where: D_T : Diameter of the storage tank; V_T : Volume of the storage tank, A_r : area of absorber;
201 CR: concentration ratio.

202 It should be noted that the concentrator should not oxidize during operation as it will reduce its
203 smoothness. In the market, polished sheets that have high reflection properties include mirrors
204 [13], steel sheets and aluminum foil. Each of the three concentrators was installed on the water
205 heater, and various experiments were carried out to ultimately select the concentrator that had the
206 best thermal performance. The general characteristics of the designed ICS SWH are summarized
207 in Table 2.

208

209 Table 2: General characteristics of the water heater

210

211 **3. Construction and assembly**

212 Based on the results obtained in the previous section, an ICS SWH with a symmetric CPC
213 concentrator was developed (Figs. 3 and 4). The structure of an ICS SWH should be light and
214 portable, and also compatible with local climatic conditions. Therefore, chipboard was selected to
215 build the structure of the SWH. As the system should be thermally insulated, the structure was
216 covered with glass wool. To ensure the water heater's screen was always perpendicular to the
217 incoming sunlight, a manual sun tracker system was placed in the back of the water heater, which
218 was able to adjust the position of the regulator to desired angles (Fig. 4). This allowed to the angle
219 of the water heater to be changed daily, monthly or seasonally, for maximum efficiency. It should
220 be noted that this angle would change with the latitude of the area where it is installed.

221

222 Fig. 3 – Construction and assembly of the ICS SWH system

223

224 4. Experimental analysis and data collection

225 4.1. Thermal performance of concentrators

226 The mean daily efficiency of the system can be determined as [43].

$$227 \quad \eta_d = Q_w / Q_R \quad (9)$$

228 where: η_d : mean daily efficiency; Q_w : heat in water storage tank; Q_R : integrated solar radiation
229 on the system aperture. The value of Q_w without any water drainage during the day is determined
230 as:

$$231 \quad Q_w = M_w C_{p,w} (T_1 - T_0) \quad (10)$$

232 where: M_w : mass of water; $C_{p,w}$: specific heat of water; T_0 : initial storage water temperature; T_1 :
233 final storage water temperature.

234 The total amount of solar radiation entering the aperture area of SWH during the day (from time
235 t_0 (7:00) to time t_1 (19:00)), is obtained by integrating the intensity field $G(t)$ [44]:

$$236 \quad Q_R = A_a \int_{t_0}^{t_1} G(t) dt \quad (11)$$

$$237 \quad G_m = (\int_{t_0}^{t_1} G(t) dt) / Dt \quad (12)$$

238 where: G_m : mean daily solar radiation intensity; Dt : time interval during daily operation.

239 Therefore, the mean daily efficiency can be determined as a function of the
240 ratio DT_{mD} / G_m ($Kw^{-1}m^2$), by second-degree polynomial fitting [45]:

241 $\eta_d = C + B(DT_{mD}/G_m) + A (DT_{mD}/G_m)^2$ (13)

242 where: $DT_{mD} = ((T_0 + T_1)/2) - T_{ma}$; T_{ma} : mean ambient temperature. The coefficient C represents
243 the mean daily efficiency of the system in the case of $(T_0 + T_1)/2 = T_{ma}$, and A, B are the thermal
244 loss parameters of the system during the daily operation [46].

245 For night time operation, the thermal losses of the system are considered to be introduced with the
246 parameter U_s , which expresses the thermal performance of the system from afternoon until the
247 following morning, when the system does not catch any solar radiation. The night-time heat loss
248 coefficient (U_s) can be calculated by the following equation [47]:

249 $U_s = (\rho C_{p,w} V_T / Dt) \ln[(T_{0m} - T_{am}) / (T_{1m} - T_{am})]$ (14)

250 where: V_T : the volume of storage tank; ρ : density of water; Dt : time interval; T_0 : mean initial
251 storage water temperature; T_1 : mean final storage water temperature; T_{am} : mean ambient
252 temperature.

253 Fig. 4: The constructed ICS SWH at the testing site in Kerman

254

255 **4.2. Measurement apparatus**

256 A k-type thermocouple was used to measure the temperature of the water, T_w , and the ambient
257 temperature, T_a , with accuracy of ± 1.5 °C for temperature range of 0-200 °C. A Kipp and Zonen
258 pyranometer (Model: CMP22) was utilised to measure the radiation input to the collector. The
259 accuracy of the pyranometer was ± 5 W/m². All the data were collected using a data logger (Model:
260 ST-8891E) and a laptop.

261

262 **4.2. Data collection and error analysis**

263 **4.3. Data collection and error analysis**

264 Following the collection of water temperature and solar radiation intensity data, the raw data was
265 converted into meaningful information using relevant graphs to analyze the rate of change and
266 observe the performance of the system in hourly manner. Linear and second-order polynomial
267 functions were fitted to the data of variations of overnight thermal loss coefficients and mean daily
268 efficiency respectively. The error of curve-fitting was determined for each case using coefficient
269 of determination (R^2) and they are presented in Tables 3, 4, 6 and 7.

270

271 **4.3. Selection and testing of the concentrator**

272 To achieve optimal performance, selecting a proper concentrator has an important effect on the
273 thermal efficiency of the solar system [48]. Experiments were carried out with the installation of
274 different concentrators on the system and the evaluation of their thermal performances. Based on
275 the available material for concentrators in the market, mirror, steel sheet, and aluminum foil were
276 selected. The first experiment was performed by installing the aluminum foil on the system and
277 testing for three consecutive days (see Fig. 5). The water temperature at the start (of all tests) was
278 21 °C. On the first day of the test, the temperature of the water inside the storage tank reached 43
279 °C while on the second and third days it reached 53 °C and 61 °C respectively. By replacing the
280 concentrator and installing the steel sheet or mirror, the temperature of the water inside the storage
281 tank increased throughout the day (see Fig. 5).

282

283 Fig. 5 - Temperature changes for three consecutive days (M_b : Mirror booster , A_f : Aluminum
284 foil, S_s : Steel sheet, T_a : ambient temperature , G : incoming solar radiation intensity)

285

286 Fig. 6 – Overnight thermal loss coefficients and mean daily efficiency (M_b : □, A_f : ○, S_s : ▲)

287

288 It can be seen from Fig. 5 that the temperature reduction during the night was the highest for the
289 mirror. The variations of overnight thermal loss coefficients and mean daily efficiency are plotted
290 in Fig. 6. By fitting a second-order polynomial function [49] to the points obtained, an equation
291 for the mean daily efficiency was obtained for each material (Table 3). According to the equations
292 in Table 3, the coefficient C of mean daily efficiency equations was highest (0.667) for the mirror,
293 and it was 0.476 for the steel sheet and 0.437 for the aluminum foil. This indicates that the output
294 of the mirror was better than the steel sheet and aluminum foil during the day.

295

296 Table 3. Equations of mean daily efficiency of concentrators

297

298 In the same way, a first-order line [49] was fitted for the obtained points of the thermal loss
299 coefficient, and the equations for the 3 concentrators are shown in Table 4. The main part of
300 equation [43] is 6.9877 for the aluminum foil, 8.0035 for the steel sheet and 11.016 for the mirror.

301 The results indicate that, although the steel sheet has the second highest efficiency among the
302 experimented materials, it has an acceptable heat loss in comparison with the mirror. In addition,

303 it is cheaper and easy to install [24]. Therefore, the steel sheet was selected in the ICS SWH system
304 for monthly experiments (see Fig. 6).

305

306 Table 4. Equations of overnight thermal loss coefficient (U_s) for different concentrators

307

308

309 **4.4. Thermal performance of the ICS SWH in 6 different months in Kerman**

310 Since the solar slope angle changes every month, the angle of the ICS SWH needs to be adjusted
311 to obtain maximum solar radiation. Table 5 provides the optimal angles for obtaining the highest
312 solar radiation for some Iranian cities [50]. According to this table, for Kerman, the optimal angle.

313

314 Table 5 - The solar slope angles in degree ($^\circ$) at different months for 6 cities in Iran [50]

315

316 changes by about 7 to 8 degrees in every month. For each month, the data of the temperature
317 changes (the ambient temperature and temperature of the water in the storage tank) and the amount
318 of radiation entering the system for three consecutive days were collected and the results are
319 plotted in Fig. 7. Then, using equations (13) and (14), the mean daily efficiency and overnight
320 thermal loss coefficients were determined at different time intervals and the results are plotted in
321 Fig. 8. The equations of mean daily efficiency and overnight thermal loss coefficient by fitting the
322 obtained points are evaluated for 6 different months (see Tables 6 & 7).

323

324 Table 6: The system's mean daily efficiency (η_d) in different months in Kerman




325

Table 7: Overnight heat loss coefficient of the system (U_s) in different months in Kerman

326

327 The first monthly test was carried out in April and according to Table 4, the ICS SWH was adjusted
328 with solar slope angle ($\beta=14.55^\circ$). The water temperature was 21°C at the start of the test and the
329 temperature of the water reached 65°C on the third day of the experiment. The recorded
330 temperature changes in April are presented in Fig. 7.

331

332 Fig. 7 – T_w , T_a  and G  for three consecutive days in 6 different months

333

334 By using the efficiency equation, the coefficients C, B, and A were determined and shown in Fig.
335 8. The coefficient C is defined as the maximum efficiency of the ICS SWH system, which
336 according to the results obtained, is 0.6557. Similarly, considering Equation 14, the heat loss
337 coefficient of ICS SWH was obtained during the night (see Fig. 8).

338

339 Fig. 8 – Mean daily efficiency (left) and heat loss coefficient (right) of ICS SHW in Kerman (6
340 mounts)

341

342 In May, with the warming of the climate and changing angle of the sun, according to Table 4 for
343 Kerman (β was set to 77.1°), the ICS SHW system was tested. The pyranometer recorded a solar
344 radiation intensity of 980 W/m^2 in May. The water temperature in the storage tank reached 72°C .

345 Fig. 7 shows the temperature changes for three consecutive days. The mean daily efficiency and
346 heat loss coefficient graphs are presented in Fig. 8, and the relevant equations are listed in Tables
347 6 and 7. The coefficient C for May is 0.6129, which is less than the value in April.

348 The third test was conducted in June at a slope angle of -4.89° (slope to the North). The
349 temperature changes on three consecutive days are shown in Fig. 7. The maximum temperature of
350 the water was measured as 73°C , which was more than for the three days of the experiments in
351 April and May.

352 For testing in July, the water heater angle was set to -2.8° . The maximum solar radiation intensity
353 that the pyranometer displayed was 1120 W/m^2 . The temperature of the water after three days of
354 testing reached 74°C , and the difference between the water temperature at the start and end of the
355 test was 52°C , which was the highest difference compared to the earlier recorded months. Looking
356 at the coefficient C in the equations (Table 7), the test carried out in July has the smallest amount
357 of heat loss. To evaluate the thermal performance of the system in August, the acceptance angle
358 of the system (α) was adjusted to 83.93° . The sun's radiation intensity declined in comparison with
359 July, and hence, this month can be considered as a turning point in terms of solar radiation. Mean
360 daily performance also rose in August, whereas it had gone down in the previous months, which
361 can also be considered as a turning point in the ICS SWH. The last test was carried out in
362 September at an angle of $\beta=26.63^\circ$. The highest temperature changes for the first, second and third
363 days of the experiment were 31°C , 13.5°C , and 21°C respectively. In September, the coefficient
364 C and the efficiency of the water heater increased (See Fig. 9).

365

366 Fig. 9 – Thermal efficiency of the ICS SWH and mean daily radiation in 6 different months in
367 Kerman, Iran

368

369 **5. Conclusion**

370 This paper presented the design, fabrication and testing of an ICS SWH system for Kerman in Iran.
371 The developed ICS SWH can be used to heat water in houses or preheat water in small to medium-
372 sized industries. The main advantages of this system, compared with the available models on the
373 market, are inexpensive materials and portability. The developed system provides an alternative
374 solution for heating water, especially in remote rural areas.

375 The tested ICS SWH showed acceptable efficiency in comparison with similar systems [40].
376 Looking at the changes in coefficient C (Table 6), it can be noted that by increasing the amount of
377 radiation entering the water heater, the thermal efficiency of the system decreases, such that the
378 highest efficiency was in April and the lowest was in July. With the distribution of radiation
379 intensity in the months of August and September, the thermal efficiency of the system increased.
380 Based on these results, the highest efficiency would be in the colder months or in colder regions.
381 This is in agreement with the results reported in [51].

382 By increasing solar radiation intensity, the capacity of energy absorption of the system decreased,
383 since a temperature gradient rise will increase the heat loss due to convective heat transfer and
384 radiation [52]. Also, the results shows that by increasing the temperature of the water in the storage
385 tank, the radiated heat loss from the system increases.

386 One of the influential parameters on the system's performance is the temperature gradient between
387 the ambient air and water inside the storage tank. The results of this study showed that by

388 increasing the ambient temperature in hot climatic conditions such as those of Kerman, the radiated
389 energy potential in the system decreases. Therefore, constant withdrawal of hot water from the
390 tank can be recommended to help increase efficiency.

391 The results of the experiments with the three common concentrators – mirror booster, steel sheet
392 and aluminum foil – showed that using mirror reflection can increase the thermal efficiency of the
393 system, but on the other hand, can lead to more thermal losses in the system. The steel sheet is the
394 optimal amongst the materials tested, as it is economically affordable, stronger and also easy to
395 install in rural areas in Kerman.

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403

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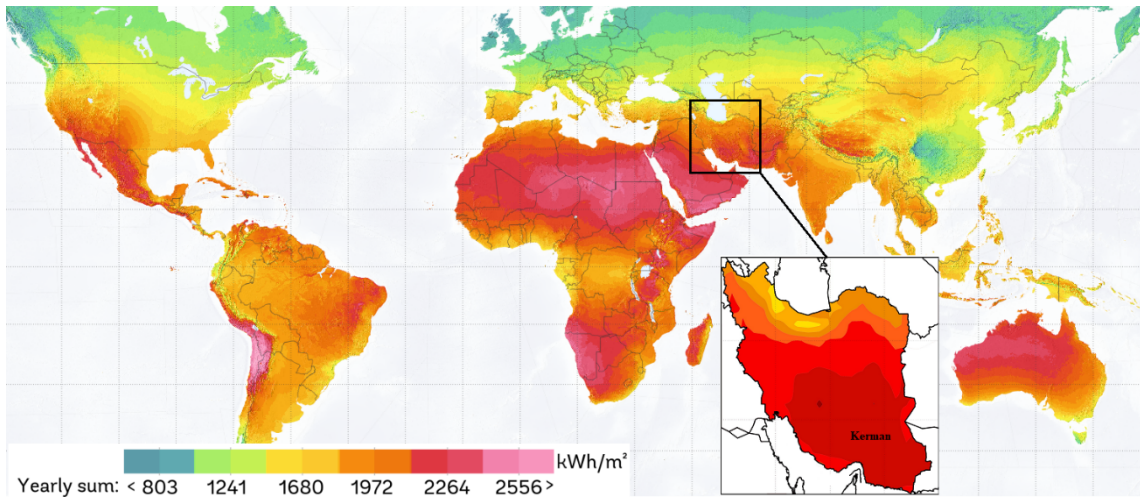
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549 Fig. 1: Comparing Annual solar radiance of Kerman with the rest of the world (kWh/m) [39].

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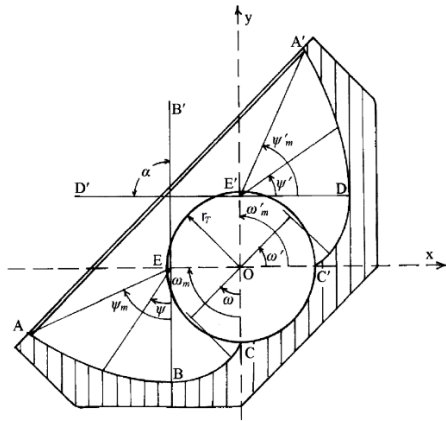
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565 Fig. 2: Cross-section view of a symmetric CPC [22]

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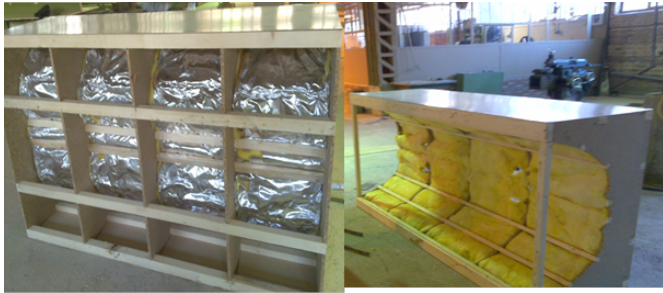
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580 Fig. 3: Construction and assembling the ICS SWH system

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Fig. 4: The constructed ICS SWH at the testing site in Kerman

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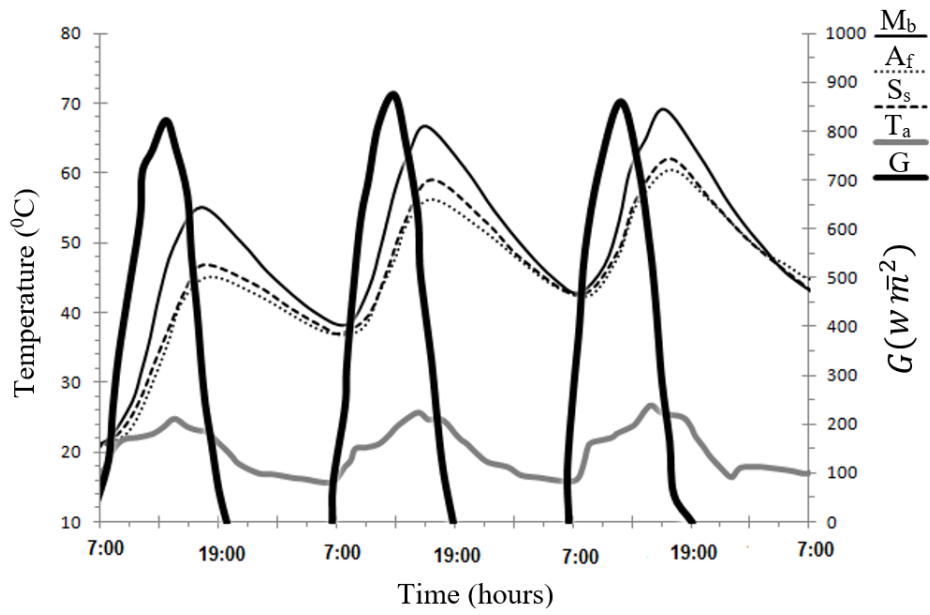
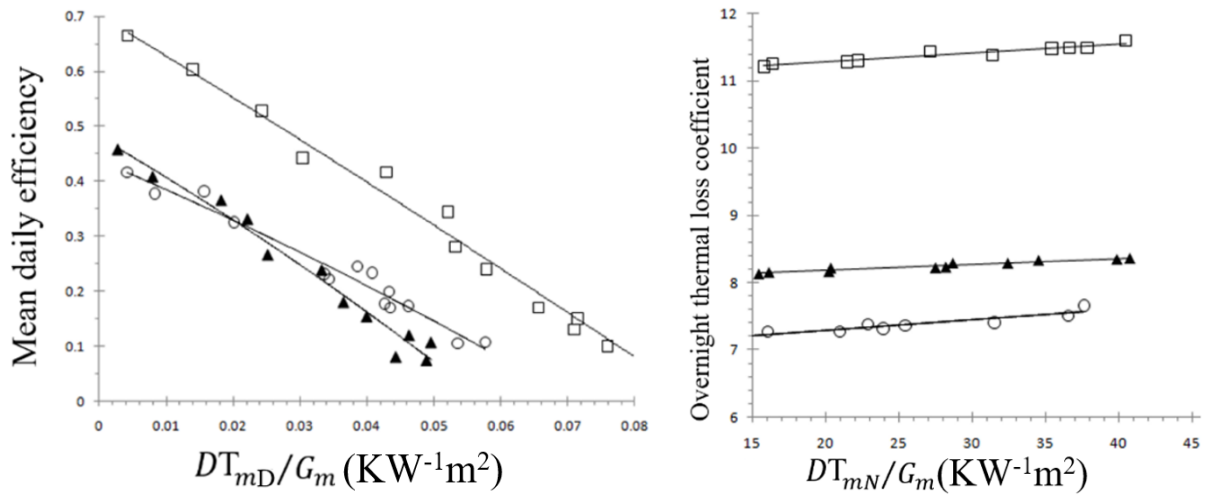


Fig. 5 - Temperature changes for three consecutive days (M_b : Mirror booster , A_f : Aluminum foil, S_s : Steel sheet, T_a : ambient temperature , G : incoming solar radiation intensity)

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Fig. 6 – Overnight thermal loss coefficients and mean daily efficiency (Mb: □, Af: ○, Ss: ▲)

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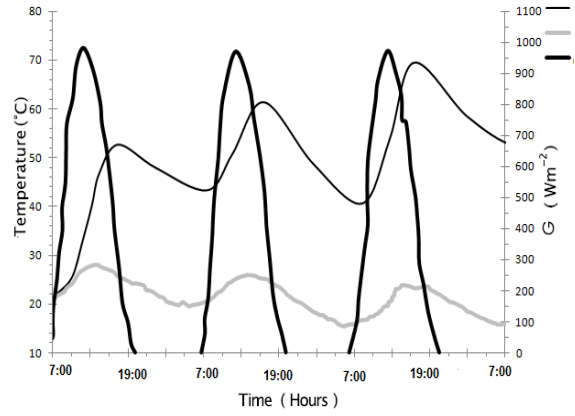
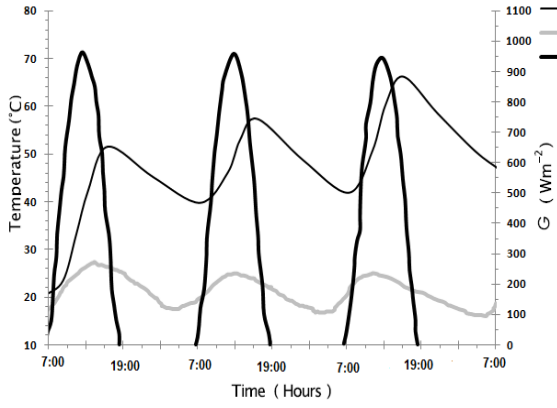
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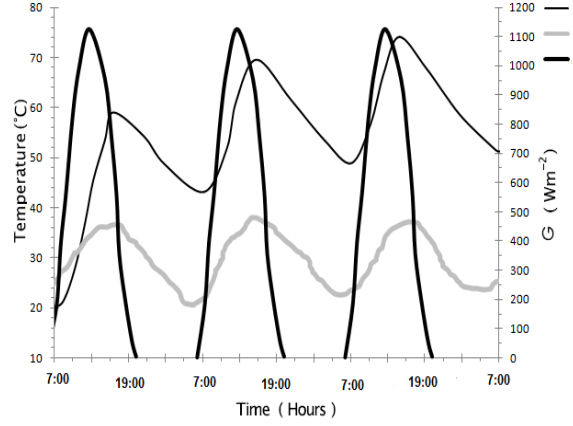
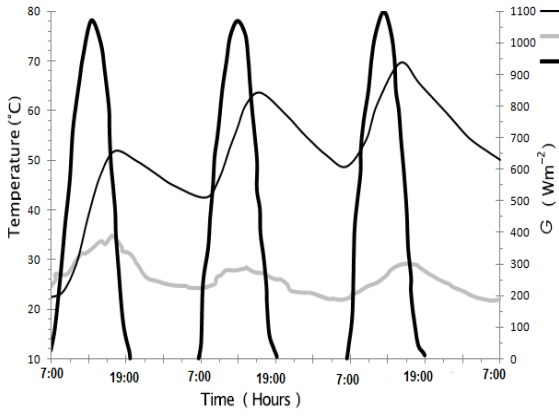
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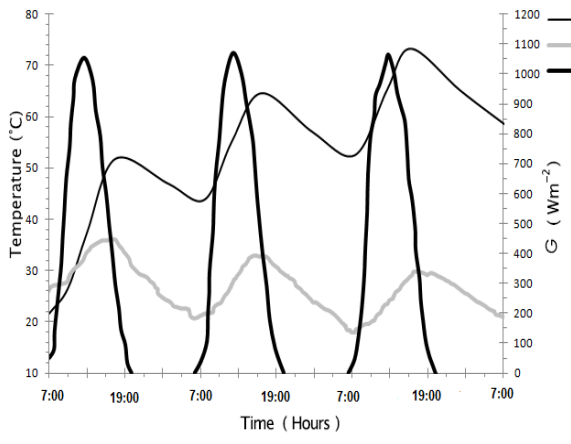
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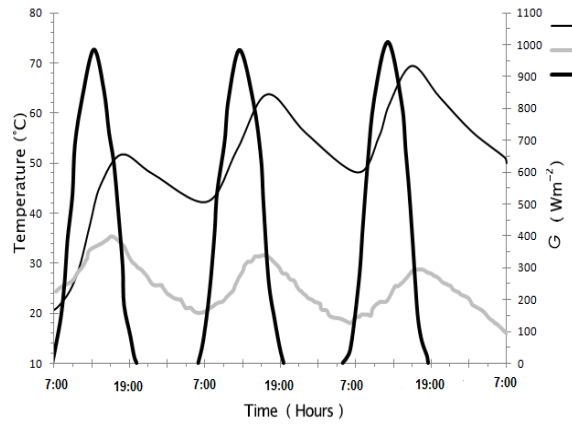


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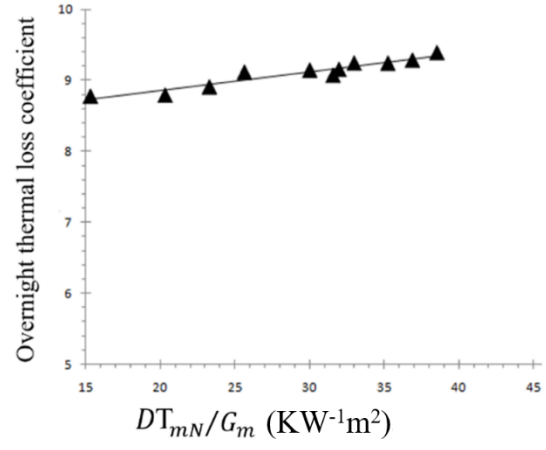
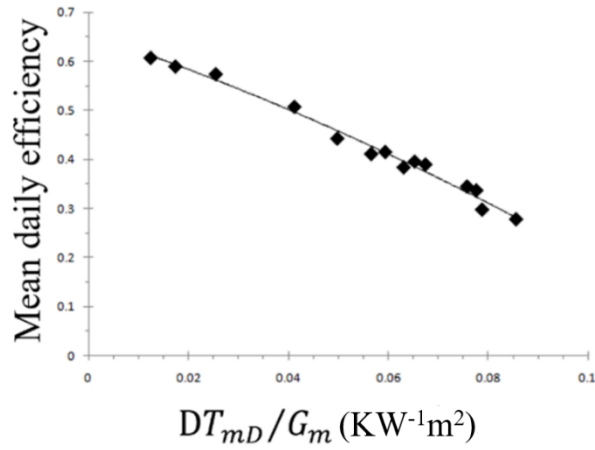


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Fig. 7 – Tw **—**, Ta **—** and G **—** for three consecutive days in 6 different months

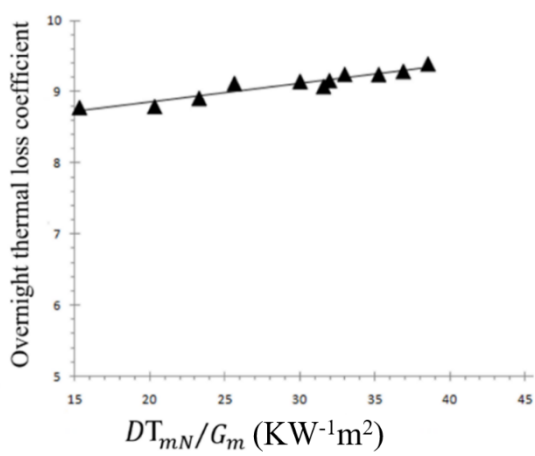
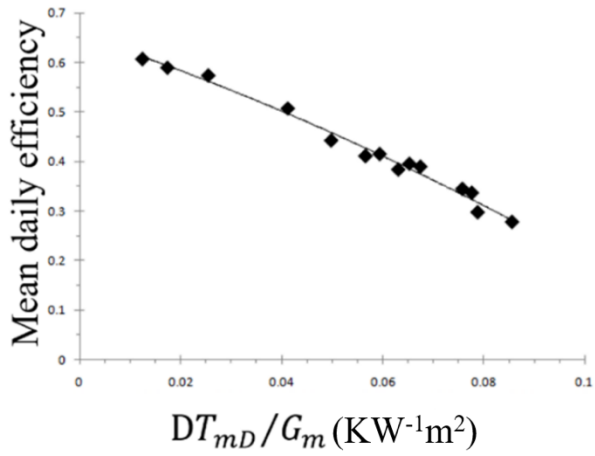


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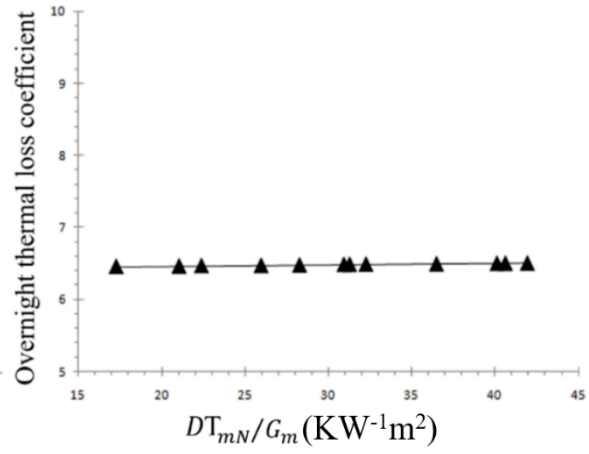
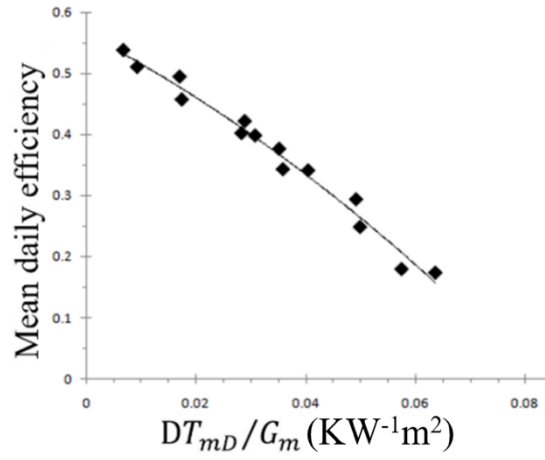


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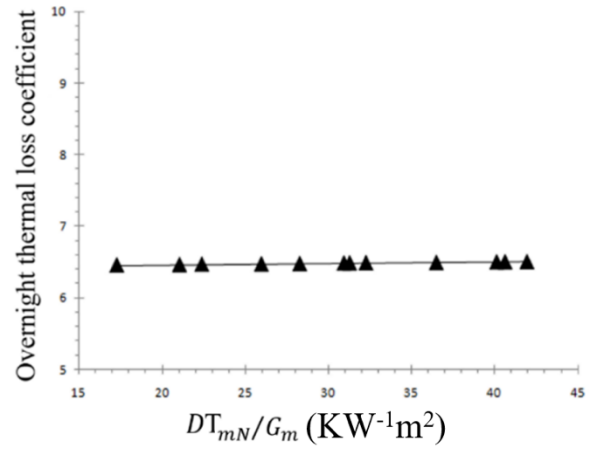
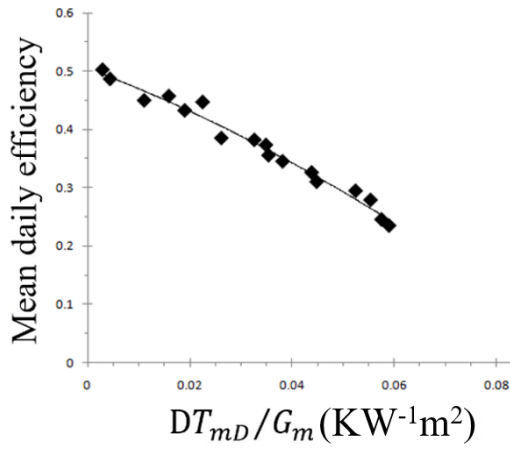
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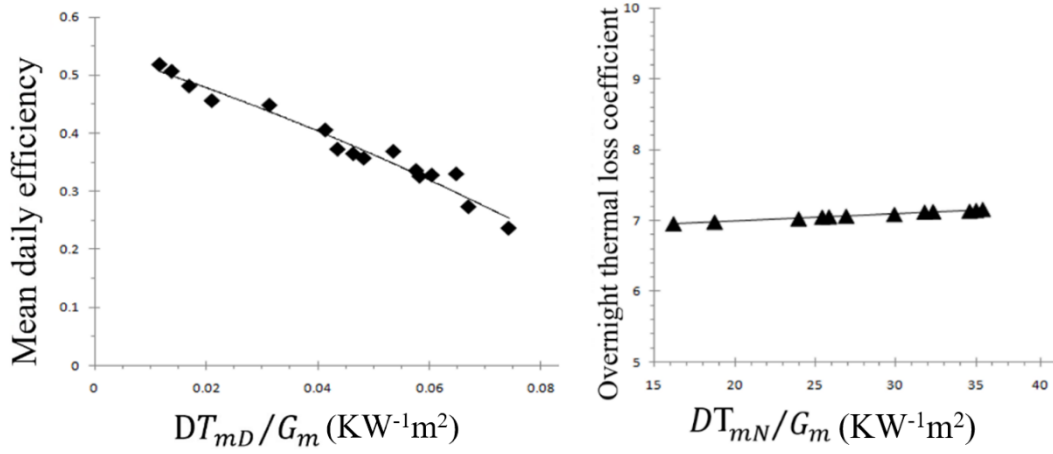
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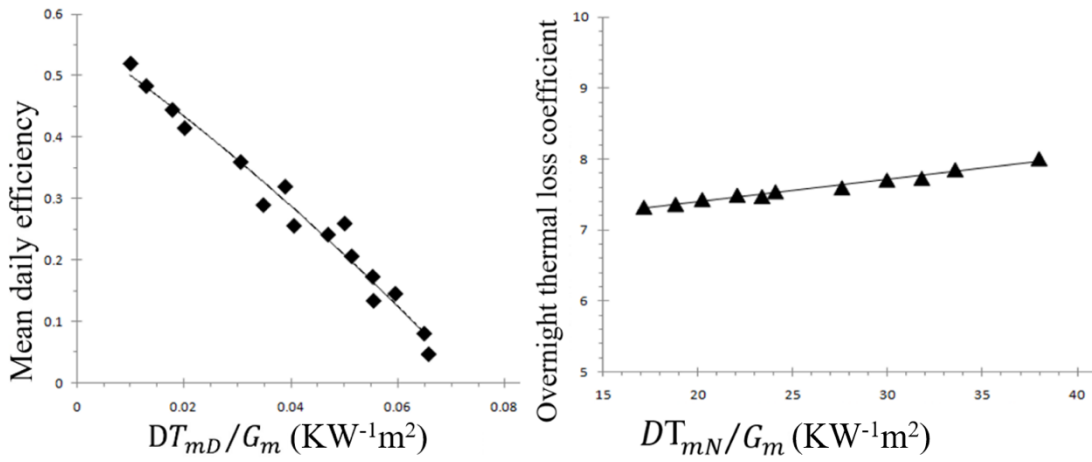
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666 Fig. 8 – Mean daily efficiency (left) and heat loss coefficient (right) of ICS SHW in Kerman (6

667 mounts)

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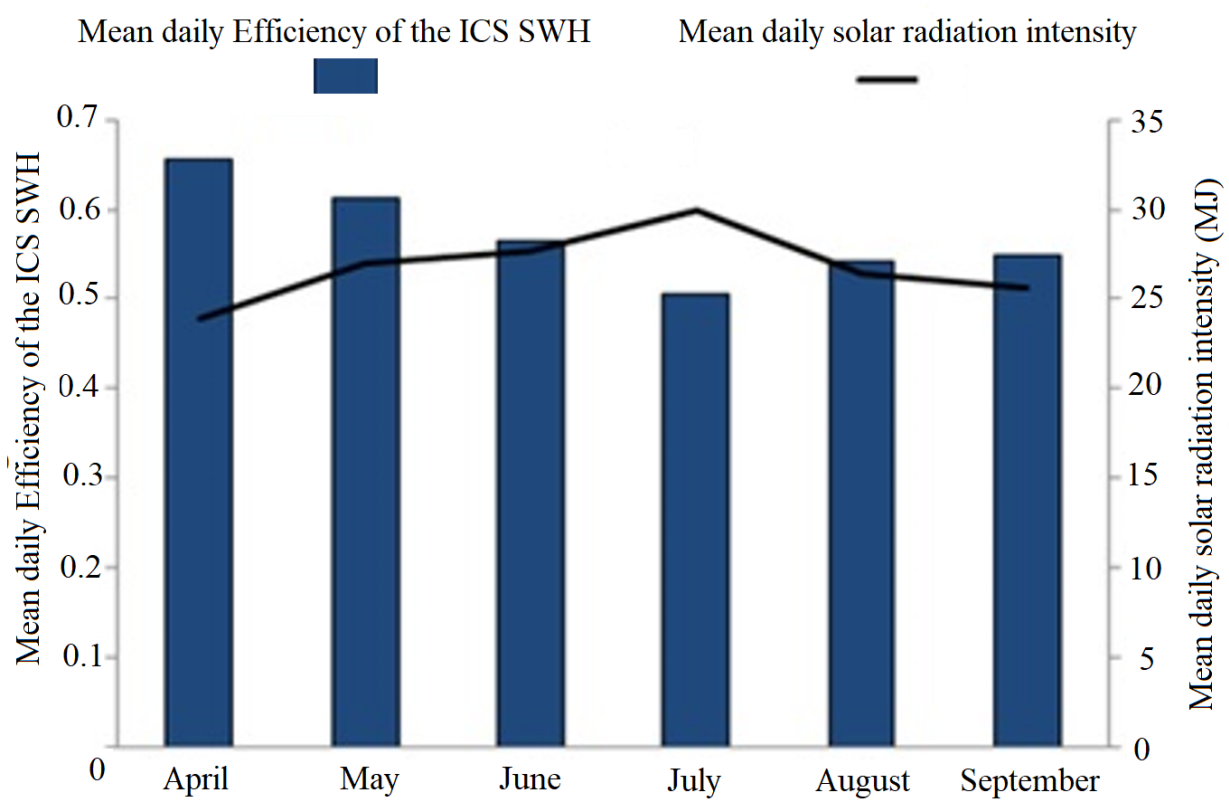


Fig. 9 – Thermal efficiency of the ICS SWH and mean daily radiation in 6 different months in Kerman, Iran

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Table 1: Configuration details of the tested ICS SWH

D_T	V_T	W_α	A_α	A_r	$\frac{V_T}{A_\alpha}$	CR
30 cm	140 liter	82 cm	16500 cm ²	14100 cm ²	84/64	1/17

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Table 2 : General characteristics of the water heater

Complete system	Dimension	200×64×113 cm
	Aperture area of systems	1.89 m ²
	Material of aperture	Simple flute glass
Concentrator	Kind of concentrator	CPC
	Material of concentrator	Steel sheet, aluminum foil, and mirror booster
Storage tank	Capacity	140 liters
	Material	Galvanized sheet
	Insulating material	Black enamel

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714 Table 3. Equations of mean daily efficiency of concentrators

Material	Mean daily efficiency equation η_d	Coefficient of determination
Mirror booster	$\eta_d = 0.667 - 7.4215(DT_{mD}/G_m) - 4.418(DT_{mD}/G_m)^2$	$R^2 = 0.9443$
Steel sheet	$\eta_d = 0.5262 - 8.2903(DT_{mD}/G_m) - 11.561(DT_{mD}/G_m)^2$	$R^2 = 0.9490$
Aluminum foil	$\eta_d = 0.4872 - 5.128(DT_{mD}/G_m) - 14.072(DT_{mD}/G_m)^2$	$R^2 = 0.9574$

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Table 4: Equations of overnight thermal loss coefficient (U_s) for different concentrators

Material	Mean daily efficiency equation η_d	Coefficient of determination
Mirror booster	$U_s = 11.016 + 0.0132(DT_{mN}/G_m)$	$R^2 = 0.8081$
Steel sheet	$U_s = 8.0035 + 0.009(DT_{mN}/G_m)$	$R^2 = 0.8742$
Aluminum foil	$U_s = 6.9877 + 0.0153(DT_{mN}/G_m)$	$R^2 = 0.7789$

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Month	Zahedan	Birjand	Shiraz	Tabas	Yazd	Kerman
January	54.14	58.37	54.64	57.69	56.72	52.83
February	44.00	47.60	40.48	47.82	47.59	2.31
March	30.01	33.28	26.22	33.07	32.50	27.83
April	14.71	17.25	13.34	17.87	16.65	14.55
May	0.97	3.89	1.31	4.68	2.98	1.77
June	-5.28	-2.80	-5.23	-1.94	-3.91	-4.89
July	-2.74	-0.10	-2.07	0.88	-0.97	-2.08
August	9.02	12.24	8.79	12.65	11.32	9.83
September	25.53	28.92	24.96	28.80	28.21	26.63
October	40.64	43.66	39.57	44.32	44.04	41.76
November	52.75	55.92	51.01	55.97	54.72	54.67
December	56.62	60.94	57.50	60.15	58.80	58.62

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Table 6: The system's mean daily efficiency (η_d) in different months in Kerman

Months	Mean daily efficiency equation η_d	Coefficient of determination
April	$\eta_d = 0.6557 - 3.3923(DT_{mD}/G_m) - 11.483(Dt_{mD}/G_m)^2$	$R^2 = 0.9760$
May	$\eta_d = 0.6129 - 4.4034(DT_{mD}/G_m) - 6.1728(DT_{mD}/G_m)^2$	$R^2 = 0.9712$
June	$\eta_d = 0.5647 - 4.6561(DT_{mD}/G_m) - 27.468(DT_{mD}/G_m)^2$	$R^2 = 0.9585$
July	$\eta_d = 0.5046 - 3.3051(DT_{mD}/G_m) - 18.606(DT_{mD}/G_m)^2$	$R^2 = 0.9643$
August	$\eta_d = 0.5429 - 2.9902(DT_{mD}/G_m) - 12.211(DT_{mD}/G_m)^2$	$R^2 = 0.9206$
September	$\eta_d = 0.5525 - 5.9753(DT_{mD}/G_m) - 22.128(DT_{mD}/G_m)^2$	$R^2 = 0.9663$

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Table 7: Overnight heat loss coefficient of the system (U_s) in different months in Kerman

Months	Mean daily efficiency equation η_d	Coefficient of determination
April	$U_s = 8.3357 + 0.0261(DT_{mN}/G_m)$	$R^2 = 0.9648$
May	$U_s = 7.1134 + 0.0223(DT_{mN}/G_m)$	$R^2 = 0.9444$
June	$U_s = 6.4208 + 0.0019(DT_{mN}/G_m)$	$R^2 = 0.8447$
July	$U_s = 5.7184 + 0.0152(DT_{mN}/G_m)$	$R^2 = 0.9259$
August	$U_s = 6.7821 + 0.0102(DT_{mN}/G_m)$	$R^2 = 0.8542$
September	$U_s = 6.79 + 0.0315(DT_{mN}/G_m)$	$R^2 = 0.9677$

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Abbreviation

D_T	Diameter of storage tank (cm)	A_α	Area of aperture (cm ²)
V_T	Volume of storage tank (liter)	A_r	Area of absorber (cm ²)
CR	Concentration Ratio	C	Maximum mean daily efficiency coefficient
ICS	Integrated collector storage	Q_w	Heat in the tank of water storage (J)
SWH	Solar water heater	Q_R	Integrated solar radiation on the aperture of the system (J)
η_d	Mean daily efficiency	T_{mN}	Mean water temperature difference during the night
r_T	Radius of storage tank	A	Acceptance angle of the system (°, rad)
DT_{mD}	Mean water temperature difference during the day	Dt_D	Time interval during daily operation
CPC	Compound-parabolic-concentrator	T_a	Ambient temperature
W_α	Width of concentrator	G	Incoming solar radiation intensity
L_T	Length of storage tank	D_t	Time interval
L_α	Length of aperture	U_s	Coefficient of thermal losses during the night
M_w	Mass of water	G_m	Mean daily solar radiation intensity
$C_{p,w}$	Specific heat of water	T_0	Initial storage water temperature
A_f	Aluminum foil	t_0	Initial time
T_1	Final storage water temperature	t_1	Final time
T_{ma}	Average ambient temperature	S_s	Steel sheet
M_b	Mirror booster	DT_{mN}	Mean water temperature difference during the night
ω	Involute concentrator's angle (°, rad)	Ψ	Parabolic concentrator's angle (°, rad)
$\acute{\omega}$	Upper involute concentrator's angle (°, rad)	$\acute{\Psi}$	Upper parabolic concentrator's angle (°, rad)

Conflict of interest: None