Magnetic Field-Sensitive Radical Pair Dynamics in Polymethylene Ether-Bridged Donor–Acceptor Systems

Hao Minh Hoang, †‡ Van Thi Bich Pham, † Günter Grampp, ‡ and Daniel R. Kattnig*¹§

¹Ho Chi Minh City University of Technology and Education, Vo Van Ngan 01, Linh Chieu Ward, Thu Duc District, Ho Chi Minh City 700000, Vietnam
²Institute of Physical and Theoretical Chemistry, Graz University of Technology, Stremayrgasse 9/I, A-8010 Graz, Austria
³Living Systems Institute and Department of Physics, University of Exeter, Stocker Road, Exeter, Devon EX4 4QD, United Kingdom

ABSTRACT: Donor–acceptor systems forming exciplexes are versatile models for the study of magnetic field effects (MFEs) on charge recombination reactions. The MFEs originate from singlet–triplet interconversion within transient radical ion pairs (RIPs), which exist in a dynamic equilibrium with the exciplexes. Here, we describe the synthesis and MFEs of the chain-linked N,N-dimethylaniline (DMA)/9-methylanthracene (MANt) donor–acceptor system MANt–(CH₂)n–O–CH₂–CH₃–DMA for n = 6, 8, 10, and 16. The MFEs are found to increase with increasing chain length. Effects as large as 37.5% have been observed for the long-chain compound with n = 16. The solvent dependence of the MFEs at magnetic field intensity 75 mT is reported. For the range of solvent static dielectric constants ε_v = 6.0–36.0, the MFEs go through a maximum for intermediate polarities, for which the direct formation of RIPs prevails and their dissociation and reencounter are balanced. Field-resolved measurements (MARY spectra) are reported for solutions in butyronitrile. The MARY spectra reveal that for n = 8, 10, 16, the average exchange interaction is negligible during the coherent lifetime of the radical pair. However, singlet–triplet dephasing broadens the lineshape; the shorter the linker, the more pronounced this effect is. For n = 6, a dip in the fluorescence intensity reveals a nonzero average exchange coupling of the order of ±5 mT. We discuss the field-dependence in the framework of the semiclassical theory taking spin-selective recombination, singlet–triplet dephasing, and exchange coupling into account. Singlet recombination rates of the order of 0.1 ns⁻¹ and various degrees of singlet–triplet dephasing govern the spin dynamics. In addition, because of a small free energy gap between the exciplex and the locally excited fluorophore quencher pair, a fully reversible interconversion between the RIP, exciplex, and locally excited fluorophore is revealed by spectrally resolved MFE measurements for the long-chain systems (n = 10, 16).

INTRODUCTION

Exciplexes are excited-state charge-transfer complexes that can be formed in photo-induced electron-transfer (PET) reactions in moderately polar solutions. Magnetic field effects (MFEs) on the emission of exciplexes have been extensively studied for many years.¹⁻¹⁶ The MFEs result from the magnetic field-dependent intersystem crossing within the fully charge-separated radical ion pairs (RIPs) in equilibrium with the exciplexes (see Figure 1). By systematically studying the MFE of the exciplex emission, details of the dynamics of the spin-correlated RIP can be revealed indirectly.¹⁷⁻¹⁹⁻²³ The MFEs are theoretically accounted for in the framework of the so-called radical pair mechanism.¹⁴⁻¹⁵,¹⁷⁻²⁰ It relies on the quantum-coherent mixing, by the hyperfine interaction (HFI) and its interplay with the Zeeman interaction, of the singlet (S) and three triplet sublevels (T₂, T₄) describing the spin configurations of the two uncompensated electron spins on the radical ions (hyperfine mechanism). Specifically, for high magnetic field intensities, the degree of singlet–triplet mixing is reduced as the magnetic field lifts the degeneracy of the triplet and singlet states (see Figure 2a), thereby reducing the number of states accessible to spin mixing. As a consequence, the singlet–triplet conversion is impeded and the yield of the charge-recombination products accessible from the singlet radical pair, such as the exciplex, increases. For moderate field intensities, the alternative Δg-mechanism is typically irrelevant as the difference in the Larmor precession frequencies of the two radicals, μ₀γB₀/h, is small compared with their inverse lifetime. Figure 1 depicts a comprehensive reaction scheme of an intramolecular PET processes in an exciplex-forming fluorophore-bridge-quencher system with flexible linker. The ordinate symbolizes free energy and the abscissa corresponds.

Received: June 4, 2018
Accepted: August 20, 2018
Published: August 31, 2018

DOI: 10.1021/acsomega.8b01232
ACS Omega 2018, 3, 10296–10305
The degree and characteristics of the magnetosensitivity of the RIP depend on the strength of the exchange interaction of the two radicals. By energetically splitting singlet from triplet states, the primary effect of this interaction is to suppress MFEs (see Figure 2b; left scenario). As the exchange integral $J(r)$ decreases approximately exponentially with the inter-radical distance $r$, $J(r) = J_0 \exp(-\xi r)$, large MFEs are often dependent on the diffusive separation of the two radicals (followed by their reencounter and recombination for product formation). Typically, $\xi \approx 0.4 \text{ nm}^{-1}$ while $J_0$ varies from system to system. Consequently, for distances of the order of tens of angstroms the exchange interaction falls off to zero, and the degeneracy of the singlet and three triplet states (in the absence of an external magnetic field) enable MFEs as described in the previous paragraph. Alternatively, if the radical pair experiences a nonzero exchange interaction, significant MFEs can still occur if the external magnetic field is able to re-establish the near-degeneracy of $T_\pi$ (or $T_\sigma$, depending on the sign of $J$) and the $S$ states, for which hyperfine-induced spin mixing is again efficient (see Figure 2b; right scenario). For charge-transfer systems studied here, $J(r)$ can be estimated from ET properties such as the charge recombination free energy, the reorganization energy, the energy of the locally excited triplet state, and the electron coupling matrix element. In this case, the distance dependence of $J(r)$, that is, the $\xi$-parameter, predominantly reflects the distance dependence of (the square of) the ET coupling matrix element. For the $N,N$-dimethylaniline (DMA)/9,10-dimethylanthracene (DMAnt)-based system studied here, $J$ is predicted to be negative (following the convention as expressed in eq 2 below), that is, the scenario as shown in Figure 2b is likely relevant.

The relative diffusive motion of the radicals plays an important role for the sizes of the MFEs of exciplexes. In several studies, the solvent environment was varied to control this factor via the static dielectric constant ($\varepsilon_s$) and the dynamic viscosity ($\eta$). In nonpolar solvents, the spin mixing and, thus, MFEs are often found to be impeded by the inability of the radical pair to separate to a distance for which $J$ is small. If, on the other hand, the solvent polarity is high, the diffusive separation might be efficient. Yet, for intermolecular systems, the MFEs may still be small if the reencounter probability of the geminate radicals is small. Consequently, maximal MFEs are often observed for intermediate dielectric constants, for which the solvent polarity is chosen to balance the radical separation and the reencounter probability. In micro-heterogeneous solvent mixtures, these parameters can in addition be controlled via the solvent structure, which can be optimized to elicit very large MFEs.

For bridged intramolecular exciplex systems, the radicals cannot separate freely and interesting MFEs are expected to emerge based on the interplay of the exchange interaction and the radical dynamics as confined by the bridge and steric requirements. In particular, by interconnecting the fluorophore and the quencher moieties with a flexible linker, the diffusive excursion of the radicals and their limiting distances and thus the average exchange integral can be controlled. By choosing a suitable chain length, MFEs that markedly exceed those in freely diffusing systems have been realized. For pyrene–(CH$_2$)$_m$–DNA with $m = 16$, MFEs of approximately 45% have been reported. For linkers with $m \leq 8$, the effect of the exchange interaction is pronounced. In contrast to
intermolecular systems and the long-chain analogues, dips in the MFE versus magnetic field-curves are then observed as a result of the S–T*: mixing process for nonzero average J (i.e., as predicted by Figure 2b; right scenario). So far, MFEs have only been observed for two classes of intramolecular exciplex systems: pyrene/DMA\textsuperscript{30–32} and phenanthrene/DMA.\textsuperscript{33–38} MFEs in nonexciplex forming intramolecular radical pair systems are comparably well studied.\textsuperscript{17,41–44}

In this work, we describe the synthesis and MFEs of a new class of exciplex-forming donor–acceptor systems of the type 9-methylnanthacene (MAnt)–(CH\textsubscript{2})\textsubscript{n}–O–CH\textsubscript{2}–CH\textsubscript{2}–DMA, which feature chain-linked MAnt and DMA. The MFEs of these compounds were studied using steady-state recordings of the exciplex emission. The chain length and solvent polarity dependence of the MFEs was investigated in homogenous systems: pyrene/DMA,\textsuperscript{30} and MAnt-8-DMA which feature chain-linked MAnt and DMA. The MFEs of these systems are comparably well studied.\textsuperscript{17,41–44}

The MFEs have been determined from steady-state measurements at B\textsubscript{0} = 75 mT and the Earth’s magnetic field (B\textsubscript{0} ≈ 0 mT) and corrected for background fluorescence as detailed in eq 8. The used PA/BN binary mixtures are isoviscous and, thus, allow a systematic assessment of the static dielectric constants of the solvents, \(\varepsilon_0\) unmitigated by effects of diffusivity. For the freely diffusing donor–acceptor pair, no MFE has been observed for \(\varepsilon_0 < 7\). For \(\varepsilon_0 > 7\), the MFE increases sharply and eventually assumes a maximum value of \(\chi_E = 14.5\%\) at \(\varepsilon_0 = 18\), followed by a monotonous decrease for \(\varepsilon_0 > 18\). For the polar solvents EtCN and AN, the emission of the intermolecular exciplex is too weak to allow a reliable quantification of the MFE. These data points have thus been omitted. For the linked systems, the onset of the MFEs is observed for \(\varepsilon_0 ≈ 10\) and the effect steadily increases with increasing dielectric constant in the PA/BN mixtures.

RESULTS AND DISCUSSION
Permittivity-Dependent MFE on Exciplex Emission. Figure 3 depicts the normalized absorption and fluorescence spectra of MAnt-n-O-2-DMA and, for comparison, the intermolecular system DMAnt/ DMA. The vibronic transitions with Gaussian band shape of the exciplex were extracted by employing a model as described in refs.\textsuperscript{9,45} The fluorescence spectrum of the locally excited fluorophore moiety and the exciplex are shaded in blue and red, respectively.

Figure 3. Absorption and emission spectra of DMAnt/DMA (top; quencher concentration: 0.06 M) and MAnt-8-O-2-DMA (bottom). A mixture of PA/BN with a dielectric constant of \(\varepsilon_0 = 12\) was used as solvent. The emissions of the locally excited fluorophore moiety and the exciplex are shaded in blue and red, respectively.

Our observations can be understood as follows: MFEs of exciplexes originate from the hyperfine-induced intersystem crossing of RIPS, which may be produced directly in a PET reaction or via dissociation of the exciplex (see Figure 1). Only if the component radicals of the RIP can separate to distances for which the exchange interaction, \(J(r)\), is negligible compared with the HFIIs or, in the case of S–T*: mixing, comparable to the Zeeman interaction, can the spin evolution proceed efficiently. The increase of the MFE with increasing \(\varepsilon_0\) is usually interpreted in terms of the accessibility of these magnetosensitive, that is, separated or elongated, conformers/ states;\textsuperscript{9–11} first, as we have shown by a detailed analysis of the time-resolved MFEs of DMAnt/DMA, the propensity for the system to directly form the exciplex from the locally excited singlet state increases with decreasing \(\varepsilon_0\) whereas the tendency of the exciplex to dissociate into ions decreases. Taken together, this gives rise to a larger partition of exciplex formation via magnetic field-independent pathways, a consequence of which is that the MFE is decreased. Second, even if RPs are formed, low permittivities/high viscosities will impede the separation of the radicals to distances that allow favorable ISC and thus large MFEs. As for the decrease of the MFE for EtCN (\(\eta = 0.39\) cP) and AcN (\(\eta = 0.34\) cP), we observe that these solvents are less viscous than the PA/BN mixtures (\(\eta = 0.58\) cP).\textsuperscript{9–11} The associated increase in mobility is, however, not expected to account for the decrease in the MFE. In fact, model calculations employing the stochastic Liouville equation to incorporate the chain dynamics and the distance dependence of the exchange interaction and back ET rate suggest that for systems for which the average exchange interaction is not substantial, the MFE increases with increasing mobility (see Figure 4 in ref 30). On the other hand, the decrease in the MFEs can be attributed to altered permittivity dependence of the n = 6 compound in detail as the MFE was found to be very small (\(\chi_E = 1.0\%\) in BN; see below).
back ET rates (see Figure 1). Further, the reencounter rate of the separated radicals could be decreased with increasing permittivity because of increased charge shielding, such that the pair loses its spin coherence (or is quenched by impurities) prior to its reencounter. In this case, the magnetosensitivity will be suppressed, as the magnetic field is too weak to impact upon the equilibrium properties. This argument is frequently employed in the context of freely diffusing radical pairs.7,12,46 Together with the increasing accessibility of the magnetosensitive conformers discussed above, a maximum in the permittivity dependence of the MFE is expected and in fact observed. The maximum MFE values in the polymethylene ether-linked systems with long chains (n = 10, 16) are larger than the freely diffusing systems.9−11 This can be predominantly attributed to an increased probability of geminate radical-pair recombination in the bridged systems. For the short-chain systems, the effects of the exchange interaction are expected to become more significant such that the MFEs are reduced despite enhanced recombination probabilities. In addition, the flexible linker is expected to impact the ratio of direct exciplex versus RIP formation.23 Magnetic field-dependent measurements to be reported below suggest that for n = 8, 10, and 16, the second argument applies exclusively.

**Magnetic Field-Dependent Measurements.** We have further studied the MFEs of the exciplexes for MAnt-n-O-2-DMA as a function of the magnetic field intensity, B₀. Figure 5 summarizes our results for neat BN. For n = 8, 10, and 16, the MFE increases monotonously with B₀, which is in line with the radical pair mechanism in the absence of a marked low-field effect. In particular, the spectral response suggests a short-lived radical pair that does not experience a significant exchange coupling during the diffusive excursions that determine the MFEs. In this scenario, the MFE encompasses the coherent mixing of the degenerate singlet (S) and triplet states (T⁺, T₀, T⁻) at B₀ ≈ 0 mT and S→T₀ conversion at high-fields, for which the T⁺ and T⁻ states are energetically detached by the Zeeman interaction. For the comparable model system of methylene-linked pyrene/DMA, 12 or 11 intervening CH₂-groups are necessary to similarly alleviate obvious extrema due to a nonzero exchange coupling in AN and diethylene glycol, respectively.26,30 As for MAnt-8-O-2-DMA comprises 11 atoms (10 methylene groups and an ether oxygen atom), the two families of compounds appear very similar. We have further extracted the B₁/₂-values, that is, the magnetic field values for which the delayed exciplex fluorescence reaches half of its saturation relative to B₀ = 0 mT; for n = 8, 10, and 16, we determine this to be 18, 9.5, and 5.6 mT, respectively. We observe that for n = 8 and 10, the B₁/₂-values are markedly larger than the corresponding value for the free fluorophore−quencher-pair DMAnt/DMA in the limit of low donor

**Figure 4.** MFEs on the exciplexes from freely diffusing DMAnt/DMA and chain-linked MAnt-n-O-2-DMA as determined from steady-state-luminescence measurements (semi-filled circles with error bars) in PA/BN mixtures of various static dielectric constants (εₛ = 6 ÷ 24.7) and neat propionitrile (εₛ = 28.3) and AN (εₛ = 36.0).

**Figure 5.** Dependence of the MFEs of the exciplex (χₑ; see eq 8) of the polymethylene-linked compounds MAnt-n-O-2-DMA with n = 6, 8, 10, and 16 on the external magnetic field intensity. The MFEs were obtained by steady-state measurements of the exciplex emission intensity at 550 nm using neat BN as solvent. The bottom panel shows the MFEs normalized by their respective value at 80 mT.
concentration ($B_{1/2} = 5.3$ mT). This can be attributed to a reduced lifetime of the radical pair and/or singlet–triplet dephasing. In fact, the lifetime effect has been suggested to lead to a broadening of the MARY lines in a study of the microheterogeneous solvation of the radical pair of DMAn/DMAn in DMSO/toluene mixtures, which likewise promote large MFEs by increasing the reencounter probability. Singlet–triplet dephasing is the consequence of the randomization of the S/T-coherences by a time-varying $\chi(r)$ as caused by the molecular motion. Given that for $n \geq 8$, no spectral fingerprints of an average exchange coupling are revealed in Figure 5, the picture of brief random encounters with large exchange coupling, interrupted by comparably long stretches of “free” evolution, emerges. We note again that the exchange interaction decays exponentially with the distance between two radicals. In the freely diffusing DMAn/DMAn system, two radicals can separate freely to the region where the exchange interaction is negligible during the majority of the coherent RP lifetime such that its effect can be neglected. The fact that for $n = 16$, the $B_{1/2}$-parameter is only marginally larger than for the freely diffusing system suggests that for this system, most of the diffusive excursion is taking place outside the mutual sphere of influence of the donor–acceptor pair, despite the linker. For the short-chain variant with $n = 6$, a completely different picture emerges: the MARY-spectrum shows a pronounced tip crossing is not present in the short-chain variant with $n = 6$.

The reduced lifetime of the radical pair and/or singlet–triplet dephasing is the consequence of the randomization of the S/T-coherences by a time-varying $\chi(r)$ as caused by the molecular motion. For these reasons, the MARY-spectrum shows a pronounced tip crossing.

We do not explicitly treat spin relaxation as these incoherent processes are not expected to be relevant on the timescale of tens of nanoseconds but we include an abstract decay time (of the order of a typical relaxation time, i.e., several hundreds of nanoseconds but we include an abstract decay time (of the order of a typical relaxation time, i.e., several hundreds of nanoseconds but we include a numerical solution which accounts for the elementary process as well as the chain dynamics controlling the approach of the reactants. Singlet–triplet dephasing (rate $k_{ST}$) is accounted for by

$$\hat{k}_{ST} \rho(t) = -\sum_{j=S,T} k_j \hat{P}_j \hat{\rho}(t) \hat{P}_j$$

where $\hat{P}_S$ and $\hat{P}_T$ are the singlet and triplet projection operator, respectively. Note that in the context of this work, $k_S$ and $k_T$ are effective rate constants that account for the elementary process as well as the chain dynamics controlling the approach of the reactants. Singlet–triplet dephasing (rate $k_{ST}$) is accounted for by

$$\hat{k}_{ST} \rho(t) = -\sum_{j=S,T} k_j \hat{P}_j \hat{\rho}(t) \hat{P}_j$$

We do not explicitly treat spin relaxation as these incoherent processes are not expected to be relevant on the timescale of tens of nanoseconds but we include an abstract decay time (of the order of a typical relaxation time, i.e., several hundreds of nanoseconds) and the order of a typical relaxation time, i.e., several hundreds of nanoseconds). However, for this reason, we here invoke the semiclassical approximation to solve eq 1 for the accumulated density operator $\hat{\rho} = \int_0^\infty dt \rho(t)$.

Details of this approach are summarized in the Supporting Information. The singlet yield is then given by

$$Y_S(B_0) = k_S \text{Tr} \left( \hat{P}_S \hat{\rho}(B_0) \right)$$

Assuming that the exciplex emission results from a field-independent contribution $\phi_0$ and a field dependent contribution related to eq 6 by $\phi(B_0)$, the MFE is eventually calculated from

$$\chi(B_0) = \frac{Y_S(B_0) - Y_S(B_0 = 0)}{Y_S(B_0 = 0) + c}$$

where $c = \phi_0/\phi_1$ is a constant larger than zero.

Using DFT-derived hyperfine coupling constants (summarized Tables S1 and S2 in the Supporting Information) and a few assumptions on the typical size of pertinent parameters,
the outlined model can account for the field-dependence of the exciplex emission of the MAnt−(CH2)n−O−CH2−CH2−DMA compounds with n ≥ 8. Figure 6 shows the fits to the experimental MFEs obtained by systematically varying the singlet recombination rate constant $k_S$, the singlet–triplet dephasing rate $k_T$, and $c$. Here, $J_{av} ≈ 0$ mT was assumed as the MARY curves do not show discernible minima that could be accounted for by S–T2-mixing. In addition, we have assumed that $k_T$ significantly exceeds $k_S$ (in practice, we used $k_T = 10k_S$). This choice can be rationalized by observing that the triplet recombination occurs in the Marcus-normal region, while the singlet recombination is expected to be relatively slow. This is attributed to the back ET being located deep in the inverted region, and the fact that the exciplex formation is tied to more stringent requirements with respect to the stacking of the reactants than simple outer-sphere ET processes. Although the data do not allow the determination of $k_T$, it is noteworthy that the choice $k_S = 10k_T$ did not provide a convincing agreement with the experimental data. Further, $\tau_{kT} = 250$ ns was used, which corresponds to the spin relaxation times expected for organic radicals in solutions. The difference in g-factors was neglected as it is too small to elicit discernible effects at the low magnetic field intensities employed here. With these choices, the MFEs can be well-modeled as is obvious from Figure 6. The fitting parameters are summarized in Table 1 and reveal a singlet recombination rate of $k_S ≈ 0.1–0.2$ ns$^{-1}$. While $k_S^{-1} ≈ 10$ ns for $n = 16$ and $n = 8$, we find $k_S^{-1} ≈ 5$ ns for $n = 10$. In addition, $c = \mu_0 / \mu_1$ is smallest for $n = 10$. The found recombination rates also justify our initial choice of refraining from an explicit modeling of spin relaxation. This is due to the modulation of (predominantly) the HFIs by rotational diffusion, which do not impact the MFEs on the short timescales relevant here. On the other hand, singlet–triplet dephasing as resulting from the modulation of the exchange coupling is found to be essential to model the width of the MARY lines. $k_T$ increases strongly with decreasing $n$. For $n = 8$, the singlet–triplet coherence dephases on the timescale of nanoseconds. At the same time, $c$ increases, suggesting that tightly bound (e.g., stacked) configurations are sampled at a much higher rate for this compound than its homologs of larger $n$. In Table 1 we also summarize the MFE for $c = 0$, that is, for the hypothetical scenario that the exciplex is exclusively formed via magnetosensitive RIPs. With regard to these data, it is interesting to note that the small MFEs for $n = 8$ are not the result of exceedingly fast singlet–triplet dephasing but a larger propensity to form the exciplex via field-independent channels.

Although this approach is obviously valid for the compounds with $n ≥ 8$, the results are less convincing for $n = 6$, as is shown in the Supporting Information (Figure S3). While the model qualitatively accounts for the dip in the MFEs once the average exchange interaction is increased to $J_{av} ≈ 5$ mT, the predicted curve levels off too fast to fully account for the experimental saturation behavior. The best agreement is found for $k_{ST} ≈ 1$ ns$^{-1}$, $k_S ≈ 0.1$ ns$^{-1}$, and $c ≈ 12$. Interestingly, increasing $k_{ST}$ decreases the agreement with the experimental data by predominantly flattening out the negative dip, rather than substantially increasing the (too small) linewidth. On the basis of this observation, we suggest that the broader magnetic field response observed for the $n = 6$ compound is likely due to strain in $J_{av}$ rather than due to increased singlet–triplet dephasing. In particular, contributions from configurations with larger exchange coupling, that is, $J_{av} ≈ 10$ mT, appear to be significant. Several interconverting populations of radical pairs with variable $J(r)$ will have to be invoked to model the experimental data well; however, this is beyond the scope of this paper. Eventually, note that, based on an analysis summarized in ref 8, the exchange interaction is expected to be negative for the studied compounds.

![Figure 6](image-url)

**Figure 6.** Simulations of the MARY curves of MAnt-n-O-2-DMA with $n = 8$ (blue), 10 (green), 16 (red) based on the semiclassical model outlined in the main text and the parameter values as reported in Table 1.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>$k_S$/s$^{-1}$</th>
<th>$k_T$/s$^{-1}$</th>
<th>$c$</th>
<th>$\chi$ for $c = 0$ (at 75 mT)</th>
</tr>
</thead>
<tbody>
<tr>
<td>$n = 16$</td>
<td>$1.0 \times 10^8$</td>
<td>$8.1 \times 10^6$</td>
<td>0.29</td>
<td>0.63</td>
</tr>
<tr>
<td>$n = 10$</td>
<td>$2.0 \times 10^8$</td>
<td>$2.1 \times 10^8$</td>
<td>0.21</td>
<td>0.25</td>
</tr>
<tr>
<td>$n = 8$</td>
<td>$1.0 \times 10^8$</td>
<td>$2.1 \times 10^9$</td>
<td>1.1</td>
<td>0.23</td>
</tr>
</tbody>
</table>

*Uncertainties: ±0.4 × 10$^8$ s$^{-1}$ for $k_S$, ±10% for the other parameters except for $k_{ST}$ at $n = 16$, for which the indicated value is an estimate of the order of magnitude.*

**Table 1. Parameters Used to Model the Magnetic Field Dependence of the Exciplex Emission Intensity of MAnt−(CH2)n−O−CH2−CH2−DMA Compounds in BN**

ACS Omega 2018, 3, 10296--10305

DOI: 10.1021/acsomega.8b01232

ACS Omega 2018, 3, 10296--10305
The change in free energy is also reflected in the emission spectra of the exciplexes, which are hypsochromically shifted as $n$ increases (see Figures S1 and S2 in the Supporting Information). Furthermore, the solvent dependence of the exciplex emission maximum shows that the degree of charge transfer is larger in the short-chain exciplexes, that is, they resemble the ion pair more.

**CONCLUSIONS**

MFEs on the intramolecular exciplex fluorescence have been studied for the new class of compounds MAnt-6,8,O,10,16,12-DMA ($n = 6, 8, 10, 16$). The effects are interpreted in terms of the hyperfine and $S/T_1$-level crossing mechanisms, in which the magnetic field affects the $S$–$T$ conversion within the RIP by altering the energy gap of singlet and triplet states. The MFE of the exciplex exhibits a strong dependence on the static dielectric constants, $\varepsilon_{\text{r}}$, of the solvent. It initially increases with increasing polarity of the solvent, attains a maximum at $\varepsilon_{\text{r}} \approx 25$, and eventually decreases for larger solvent polarity. The solvent dependence of the MFE on the exciplex was discussed based on the effect of the solvent polarity on the separation and reencounter dynamics of the RIP and the partitioning of the quenching pathways giving way to the exciplex and the RIP, respectively. The size of the MFEs correlate strongly with the edge-to-edge distance ($r$) of the two radicals. MFEs on the intramolecular exciplexes have been shown to be significant for the long-chain systems ($n = 10, 16$). This observation can be attributed to the fully reversible interconversion between RIP, exciplex, and locally excited fluorophore due to a small free energy gap between the exciplex and the excited fluorophore.
quencher pair. Our result suggests that the driving force of exciplex formation from the locally excited fluorophore quencher pair increases with decreasing chain length.

# EXPERIMENTAL SECTION

## Synthesis of Chain-Linked Fluorophore–Quencher Systems

Exemplary procedure for 2-[4-((N,N-dimethylamino)phenyl)ethyl]-10-[9-(10-methyl)anthryl]decane (MAnt-10-O-2- DMA): 1-bromo-10-[9-(10-methyl)anthryl]decane (MAnt-10-Br, 0.65 g, 1.88 mmole) and 2-[4-(dimethylamino)phenyl]ethanol (0.56 g, 3.39 mmole) were added to the solution of 0.57 g NaH (60% oil suspension) in N,N-dimethylacetamide (15 mL). The mixture was stirred for 4 h in an ice bath at 273 K. The resulting mixture was extracted with benzene (3 × 15 mL). The combined benzene phases were washed by cold water (3 × 20 mL) and dried over MgSO4. The solvent was removed in vacuo. The product was isolated by column chromatography on silica gel with n-hexane and an n-hexane/ethyl acetate mixture (95:5) as the eluting solvent to obtain MAnt-10-O-2-DMA (0.25 g, 38% yield).


## Solvents and Sample Preparation

Solvent mixtures of PA (Aldrich 99.9%, distilled under reduced pressure, εs = 6) and BN (BN, Fluka 99%, distilled under reduced pressure, εs = 24.7) allow for a systematic variation of the dielectric constant εs in the range from 6 to 24.7 (295 K). Mixtures of different εs values were prepared observing that εs = wPAεPA + (1 − wPA)εBN where wPA is the weight fraction of PA. For these mixtures, the viscosity (η) and the refractive index (n) are almost independent of solvent composition. Furthermore, the Pekar factor (1/n2 − 1/εs) = 0.456 for εs = 15, which governs the outer-sphere ET reorganization energy and thus the rate of ET processes, varies by only ±5% in the studied εs range.

In addition, neat propionitrile (EtCN, Aldrich 99.5%, distilled under reduced pressure, εs = 28.3) and AN (Aldrich 99.8%, distilled under reduced pressure, εs = 36.0) are used to extend the range of solvent polarities. The concentration of ether-linked fluorophore–quencher pairs was 2 × 10⁻⁵ M. Samples were prepared in septa-sealed quartz cuvettes. To remove dissolved oxygen, all solutions were bubbled with nitrogen for 15 min prior to measurements.

## Steady-State MFE Measurements

The MFEs on the intramolecular exciplex emission were obtained by using the experimental setup described in refs. 9,11. The exciplex luminescence intensity was recorded at 550 nm, whereas the fluorophore moiety was excited continuously at 374 nm. The fluorescence intensity was sampled three times in the absence and presence of an additional external magnetic field; the two conditions were alternated and each was acquired for 60 s. All measurements were conducted at 295 K. Fluorescence signals have been background-corrected. The three repetitions were analyzed independently and the experimental errors were estimated according to the method described in refs. 9,11. The MFE is defined as the relative change of the fluorescence intensity of the exciplex, I0(λem, B0), in the presence of an external magnetic field relative to the zero-field scenario. It was evaluated from

$$\frac{\Delta I}{I_0} = \frac{I(\lambda_{em}, B0) - I(\lambda_{em}, 0)}{I(\lambda_{em}, 0)}$$

Here, I(λem, B0) and I(λem, 0) are the mean intensities sampled over the 60 s intervals at λem in a saturating magnetic field (B0) and in the absence (B0 = 0) of an additional external magnetic field, respectively. The MFE is the residual emission of the locally excited fluorophore at the emission wavelength λem in the absence of quencher. I0 and I1 are the emission intensities of the locally excited fluorophore in the presence and absence of the quencher, respectively, which have been obtained from the decomposition of the spectrally resolved fluorescence spectra in contributions from the fluorophore and the exciplex. B0(λem) is the mean background intensity recorded for a sample of only the solvent. The subtractive term in the denominator accounts for the residual fluorescence of the locally excited fluorophore at the wavelength used to monitor the exciplex emission. I0 has been determined from spectra of the dimethyl anthracene recorded under the same experimental conditions and for the same concentration as used in the actual experiments. This correction is crucial for intermolecular exciplexes in polar solutions, for which the exciplex emission is small.

## ASSOCIATED CONTENT

# Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsomega.8b01232.

Extracted exciplex spectra, solvent dependence of the maxima of exciplex emission, solution of eq 1 in the semiclassical limit, DFT-derived hyperfine coupling constants, tentative simulations of the MARY curve of MAnt-6-O-2-DMA, and detailed description of the synthesis of the polymethylene ether-linked donor–acceptor compounds (PDF)

# AUTHOR INFORMATION

Corresponding Author

*E-mail: d.r.kattning@exeter.ac.uk. Phone: +44 (0) 1392 72 7479.

ORCID

Daniel R. Kattning: 0000-0003-4236-2627

Notes

The authors declare no competing financial interest.

# ACKNOWLEDGMENTS

The financial support from The Royal Society (RG170378), Ho Chi Minh City University of Technology and Education (T2018-36TD) and Austrian Science Fund (FWF-Project P
21S18-N19) is gratefully acknowledged. D.R.K. is thankful to NVIDIA for supporting this research through their GPU Grant Program.

REFERENCES


