| 1  | <b>Title:</b> Lessons from two high CO <sub>2</sub> worlds – future oceans                      |
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| 2  | and intensive aquaculture   |
| 3  | Running head: Lessons from two high CO <sub>2</sub> worlds                                      |
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### 21 Abstract

Exponentially rising CO<sub>2</sub> (currently ~400 µatm) is driving climate change, and causing 22 acidification of both marine and freshwater environments. Physiologists have long known 23 that CO<sub>2</sub> directly affects acid-base and ion regulation, respiratory function, and aerobic 24 performance in aquatic animals. More recently, many studies have demonstrated that elevated 25 CO<sub>2</sub> projected for end of this century (e.g. 800-1,000 µatm) can also impact physiology, and 26 27 have substantial effects on behaviours linked to sensory stimuli (smell, hearing and vision) both having negative implications for fitness and survival. In contrast, the aquaculture 28 industry were farming aquatic animals at CO<sub>2</sub> levels that far exceed end of century climate 29 change projections (sometimes >10,000 µatm) long before the term "ocean acidification" was 30 coined, with limited detrimental effects reported. It is therefore vital to understand the 31 32 reasons behind this apparent discrepancy. Potential explanations include: 1) the use of "control" CO<sub>2</sub> levels in aquaculture studies that go beyond 2100 projections in an ocean 33 34 acidification context; 2) the relatively benign environment in aquaculture (abundant food, 35 disease protection, absence of predators) compared to the wild; 3) aquaculture species having been chosen due to their natural tolerance to the intensive conditions, including CO<sub>2</sub> levels; 36 or 4) the breeding of species within intensive aquaculture having further selected traits that 37 38 confer tolerance to elevated CO<sub>2</sub>. We highlight this issue and outline the insights that climate change and aquaculture science can offer for both marine and freshwater settings. Integrating 39 these two fields will stimulate discussion on the direction of future cross-disciplinary 40 research. In doing so this article aims to optimise future research efforts and elucidate 41 effective mitigation strategies for managing the negative impacts of elevated CO<sub>2</sub> on future 42 43 aquatic ecosystems and the sustainability of fish and shellfish aquaculture.

#### 44 Introduction - Climate change, high CO<sub>2</sub> and global food security

In 2015 atmospheric CO<sub>2</sub> concentrations had risen to an annual average higher than 400 45 µatm the first time in over 800,000 years (Lüthi et al., 2008, Dlugokencky & Pieter, 2016), as 46 a result of anthropogenic CO<sub>2</sub> emissions. The potential implications of this post-industrial rise 47 in CO<sub>2</sub> were predicted over 110 years ago (Krogh, 1904), yet it was only recently that 48 governments agreed to take action on this issue. Despite 196 nations taking an unprecedented 49 50 stance on climate change last year by signing the COP21 agreement to curtail emissions, CO2 concentrations are still projected to approach 1000 µatm by 2100 (Pörtner et al., 2014). Around 51 a quarter of anthropogenic CO<sub>2</sub> emissions have been absorbed by the oceans (Pörtner et al., 52 53 2014). Whilst this results in a phenomenon commonly referred to as ocean acidification, elevated atmospheric CO<sub>2</sub> is also driving a large elevation in the average aquatic CO<sub>2</sub> in fresh 54 55 and brackish water systems, regardless of diurnal and seasonal variation. What's more, seasonal oscillations of aquatic CO<sub>2</sub> in the future are predicted to amplify over time which will 56 57 likely result in CO<sub>2</sub> levels that exceed 1000 µatm for several months each year well before 58 2100 (McNeil & Sasse, 2016). Occurring simultaneously with warming, pollution, habitat degradation, disease outbreaks and overfishing, this aquatic acidification is therefore 59 threatening not only aquatic ecosystems but also global food security (FAO, 2014, Porter et 60 61 al., 2014).

Anthropogenic CO<sub>2</sub> emissions accelerate alongside growth of the global human population, which is projected to exceed 9.6 billion by 2100 (Gerland *et al.*, 2014). This same growth has also resulted in at least 80% of world fish stocks being overexploited (FAO, 2014, Pauly & Zeller, 2016). Aquaculture is therefore crucial to ensure the continued provision of fish and shellfish protein for human consumption, particularly for developing countries and small island nations (Bennett *et al.*, 2016). Indeed, aquaculture is one of the fastest growing

food producing industries globally (8.8 % annual growth for the last 30 years) (FAO, 2014), 68 and it is the only foreseeable way of increasing seafood<sup>†</sup> production in the face of this human 69 population expansion. However, to ensure aquaculture is able to maximise its potential for 70 71 addressing global food security a number of challenges need to be resolved concerning water availability and quality, environmental impacts, and vulnerability to changing climatic 72 conditions. Recirculating Aquaculture Systems (RAS) address many of these issues (Martins 73 74 et al., 2010) and enable the sustainable intensification of aquaculture. These systems significantly reduce water requirements, relocate production of aquatic organisms away from 75 76 a natural environmental setting and minimise environmental impacts. They also enable a tighter control of pathogens and other environmental parameters, potentially improving animal 77 welfare and biosecurity, but they create some additional problems, particularly associated with 78 79 accumulation of CO<sub>2</sub>.

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### • A common problem, two perspectives

Physiologists have known for decades that raising the CO<sub>2</sub> partial pressure in water to 81 well above atmospheric levels (e.g. 10,000 µatm) has a direct effect on aquatic organisms in 82 terms of acid-base and ion regulation, respiratory function, and aerobic performance (Cameron 83 & Randall, 1972). More recently, climate change studies have shown that CO<sub>2</sub> levels projected 84 85 for end of this century (e.g. 800-1,000 µatm), can negatively affect development, physiology and fitness-related behaviours in aquatic animals (see below). Due to the very high stocking 86 densities achieved in most aquaculture settings, as well as the methods employed to control pH 87 and O<sub>2</sub>, CO<sub>2</sub> often accumulates, particularly in RAS. However, despite recent evidence on the 88 potential detrimental effects of CO<sub>2</sub> exposure at a level projected for 2100 (1,000 µatm), the 89 aquaculture industry was intensively farming fish and shellfish successfully at much higher 90

<sup>&</sup>lt;sup>†</sup> Seafood in this context refers to all fish and shellfish species produced under fresh, brackish or marine conditions and intended for human consumption

91 CO<sub>2</sub> levels long before the term "ocean acidification" was coined. The levels at which the 92 effects of CO<sub>2</sub> are perceived as problematic therefore appear to differ greatly between the 93 connected yet traditionally disparate fields of climate change and aquaculture (Fig 1.).

94 Current guidelines for intensive RAS propose safe CO<sub>2</sub> levels ranging from 15 to 40 mg/l (Fivelstad et al., 1999, Blancheton, 2000, Petochi et al., 2011, Fivelstad et al., 2015). 95 These equate to an upper limit of  $CO_2$  ranging from >5,000 to >30,000 µatm which are 12.5 to 96 97 75 times higher than current atmospheric levels respectively. Furthermore, far from being an issue exclusively associated with RAS and finfish production, elevated CO<sub>2</sub> levels appear 98 synonymous with intensive aquaculture more generally. For example, over 40 % of Norwegian 99 100 salmon smolt hatcheries (flow-through and RAS) report CO<sub>2</sub> levels >5,400 µatm (Noble et al., 101 2012), whereas Bangladeshi shrimp ponds are shown to experience  $CO_2$  levels averaging >17,000 µatm (Saksena et al., 2006, Sahu et al., 2013). 102

In stark contrast, recent studies emerging from aquatic acidification research have 103 104 demonstrated that just 2 to 2.5-fold increases in CO<sub>2</sub> levels projected for the end of this century 105 (e.g. 800-1,000 µatm) can have dramatic and long-lasting effects on the development, physiology and behaviour of both fish and invertebrates (Briffa et al., 2012, Schalkhausser et 106 al., 2012, Heuer & Grosell, 2014, Watson et al., 2014, Welch et al., 2014). For example, 107 exposure to 1,000 µatm during early life cycle stages has been shown to result in reduced 108 survival as well as a number of sub-lethal effects including tissue damage (e.g. Frommel et al., 109 2012, Chambers et al., 2014, Frommel et al., 2014), altered calcification (e.g. Arnold et al., 110 2009, Maneja et al., 2013), reduced size (e.g. Talmage & Gobler, 2009, Maneja et al., 2014), 111 reduced metabolic rate (e.g. Small et al., 2016), delayed development and altered gene 112 expression (e.g. Tseng et al., 2013, Goncalves et al., 2016) in a range of different marine 113 organisms. What is more, similar effects are also demonstrated in freshwater, with Ou et al. 114 115 (2015) showing a significant effect of elevated CO<sub>2</sub> (1,000 – 2,000  $\mu$ atm) on the larval development of pink salmon *Oncorhynchus gorbuscha*. The authors reported a reduction in
larval length, total wet and dry mass and reduced production efficiency (conversion of yolk
into tissue growth).

In impacting a diverse array of aquatic organisms during early life stages, increased 119 partial pressure of CO<sub>2</sub> in aquatic environments above present day atmospheric levels is a likely 120 bottleneck for organism production. This in turn would significantly impact aquaculture 121 122 practices that depend upon a reliable source of larvae or juveniles. In 2007, these impacts were realised with the upwelling of elevated CO<sub>2</sub>, aragonite under-saturated seawater off the US 123 west coast, significantly impacting oyster hatchery production as a direct result of changing 124 125 climatic conditions (Barton et al., 2012). In addition to providing a case study in which to investigate the impact of ocean acidification on shellfish production globally, this event 126 highlighted the significant advances achieved when climate change scientists and aquaculture 127 practitioners work closely together. Unifying their research efforts to overcome this 128 phenomenon, the climate change community and shellfish growers were able to successfully 129 identify the root cause of this issue and put in place a number of mitigation strategies and 130 monitoring protocols to minimise impacts in the future (Barton et al., 2015). 131

Far from being restricted to early life stages, a growing number of studies have also 132 shown sub-lethal physiological impacts of elevated CO<sub>2</sub> (range  $1,000 - 2,000 \mu atm$ ) in a 133 number of species which include impacted respiratory gas transport, acid-base balance and gut 134 carbonate excretion (e.g. Lannig et al., 2010, Esbaugh et al., 2012, Heuer et al., 2012, Wei et 135 al., 2015, Esbaugh et al., 2016). Rapid and efficient acid-base compensation has been 136 demonstrated in a number of species at elevated CO<sub>2</sub> concentrations (e.g. Melzner *et al.*, 2009, 137 Ern & Esbaugh, 2016, Lewis et al., 2016). However, such physiological responses incur 138 energetic costs and could therefore have negative implications for production efficiency and 139 140 body condition both in aquaculture and natural settings. Likewise, a wide range of behaviours

are shown to be disrupted under elevated CO<sub>2</sub>, such as those linked to sensory stimuli 141 (including smell, hearing and vision; e.g. Simpson et al., 2011, Nilsson et al., 2012, Roggatz 142 et al., 2016) and cognitive-related functions (such as lateralisation, learning, bold-shy 143 phenotypes and escape behaviour; e.g. Schalkhausser et al., 2012, Jutfelt et al., 2013, Hamilton 144 et al., 2014, Watson et al., 2014), which will have clear detrimental implications at the 145 population level (Munday et al., 2009, Munday et al., 2010, Chivers et al., 2014). However, 146 147 animals reared in many aquaculture settings are living in a relatively benign environment, being provided with abundant food, relatively constant environmental conditions, protection against 148 149 disease and absence of a predation threat. Therefore it is perhaps not surprising that the ecologically-relevant physiological and behavioural disruptions caused by end of century CO<sub>2</sub> 150 levels in OA studies have not emerged from aquaculture studies. Equally it may be possible 151 152 these behavioural effects have not been noted as they are not typically measured in aquaculture studies. Nevertheless, this does not mean that animals reared in an aquaculture setting are not 153 facing problems associated with elevated CO<sub>2</sub> that potentially influence their health and/or 154 production efficiency. 155

#### 156 **Cross-discipline interaction to improve understanding of CO<sub>2</sub> consequences**

Given these contrasting views, combining the knowledge that has arisen from climate 157 change and aquaculture research is crucial to allow a more in-depth understanding of the 158 physiological and ecological responses of aquatic animals to elevated CO<sub>2</sub>. The opportunity to 159 160 compare these two fields directly is appealing, and should enable a more accurate prediction of the consequences of changing climatic conditions for wild populations and intensive 161 aquaculture practices alike. However, at present such comparison is not straightforward. This 162 is partly due to the different experimental measures and reporting protocols typically adopted 163 164 by each of these scientific fields. To facilitate this process it would be fruitful to develop a 165 collective research agenda and implement standard operating procedures with respect to166 hypothesis development, experimental outcomes and data reporting.

The comparison is also complicated by rather different species often being used in 167 aquaculture compared to OA research, with the former inevitably relying on species that are 168 amenable to domestication, which may go hand-in-hand with greater environmental tolerance. 169 170 Indeed, when considering contrasting results from aquatic acidification and aquaculture fields 171 it is worth noting that responses from even closely related species can often vary significantly. For example, Ferrari et al. (2011) demonstrated a striking and unexpected difference for the 172 impact of CO<sub>2</sub> on the antipredator response of closely related damselfish species. Similarly, 173 174 Lefevre (2016) and Heuer and Grosell (2014) highlight heterogeneity in physiological responses to elevated CO<sub>2</sub> that argues against a unifying physiological theory for defining CO<sub>2</sub> 175 tolerance, and which needs to be accounted for when modelling and predicting the impacts of 176 177 climate change. Indeed explaining such interspecies variability with respect to CO<sub>2</sub> tolerance may provide a mechanistic understanding of why species used in aquaculture may be relatively 178 179 tolerant to the CO<sub>2</sub> levels prevalent within intensive production. However, it is important to note that even cod reared under end of century CO<sub>2</sub> levels (1,000 µatm) exhibit avoidance 180 behaviour towards these conditions when presented with a choice, indicating negligible 181 182 habituation and suggesting these conditions are unfavourable (Jutfelt & Hedgärde, 2013). Furthermore, a growing body of evidence shows that levels of CO<sub>2</sub> experienced in aquaculture 183 may be more detrimental than traditionally perceived (Heuer & Grosell, 2014). For example 184 185 Tirsgaard et al. (2015) and Ou et al. (2015) demonstrated detrimental effects of elevated CO<sub>2</sub> in cod and salmon respectively, species traditionally grown successfully under aquaculture 186 settings. Exposure to 9,200 µatm resulted in longer meal processing time and less efficient 187 digestion in cod (Tirsgaard et al., 2015), whilst exposure to 2,000 µatm reduced growth and 188 production efficiency in salmon larvae (Ou et al., 2015), end-point measures that are of specific 189

importance to aquaculture production. Thus differences between these two fields in the 190 perceived impact of elevated CO<sub>2</sub> cannot be explained solely by variability in interspecific 191 responses. Measuring the impact of elevated CO<sub>2</sub> on a diverse array of physiological and 192 193 behavioural endpoints, not just those traditionally perceived as important for aquaculture production, is thus vital. It is also crucial to measure these responses in as many species as 194 possible, both finfish and shellfish, as well as those traditionally perceived as CO<sub>2</sub> tolerant and 195 196 CO<sub>2</sub> sensitive. By doing so it will be possible to optimise water quality parameters within aquaculture, based on a species specific suite of physiological and behavioural CO<sub>2</sub> tolerance 197 198 endpoints. Targeting these conditions has the potential to maximise growth efficiency and health of aquaculture species, enhancing the sustainability of seafood production. With that 199 aim, it is critical to understand the practical considerations of reducing and maintaining 200 201 environmental conditions, particularly CO<sub>2</sub>, in an aquaculture context. Targets should thus be 202 set that optimise productivity and welfare of the aquaculture species, but which are equally achievable in a practical and economical context (Noble et al., 2012). 203

204 In order to optimise research efforts and ensure data are both scientifically robust and comparable, a unified protocol for selecting, manipulating, measuring and finally reporting 205 carbonate chemistry parameters is also needed. This is of particular importance given the 206 207 methods of carbonate chemistry manipulation employed within intensive aquaculture, for example the addition of a strong alkali to buffer changes in pH, such as sodium hydroxide 208 (NaOH), sodium bicarbonate (NaHCO<sub>3</sub>), calcium hydroxide (Ca(OH)<sub>2</sub>) or calcium oxide 209 210 (CaO). This is the most commonly used of all water chemistry quality management practices in aquaculture, being typically employed in a diverse array of aquaculture settings (Boyd et al., 211 2016). However, this method of pH compensation additionally elevates alkalinity, often 212 significantly beyond any natural analogue (Ellis et al., in preparation), and depending on the 213 alkali used can have dramatic indirect effects on additional water chemistry parameters, some 214

of which are shown themselves to influence a number physiological processes in aquatic 215 organisms (Boyd et al., 2016, Middlemiss et al., 2016). A further crucial issue is the selection 216 217 of experimental controls representing present day  $CO_2$  levels (400  $\mu$  atm), and we propose this should be a common reference point for both climate change and aquaculture researchers. 218 Control levels employed within aquaculture research typically exceed 1,000 µatm (range 219 1,000–3,000 µatm) (Fivelstad et al., 1999, Petochi et al., 2011), and thus surpass most of the 220 "high CO<sub>2</sub>" treatments used as end of century projections in climate change studies. To 221 complicate matters further, reporting CO<sub>2</sub> levels as mg/l in aquaculture studies overlooks the 222 223 impact of temperature and salinity on the solubility of CO<sub>2</sub> and the resulting impact these have on the partial pressure of this gas (Weiss, 1974, Dickson, 2011). For example, for the same 224 mg/l concentration the actual partial pressure of CO<sub>2</sub> varies by more than 3-fold between cold 225 226 freshwater and warm seawater (Fig. 2). This is critical because it is partial pressure (not the mg/l concentration) that determines the internal (blood) levels of CO<sub>2</sub> and its impact on 227 physiology, behaviour, growth etc. At present, the scarcity of sufficient water chemistry 228 parameters being presented, the lack of environmentally relevant controls, and the prevalence 229 of reporting CO<sub>2</sub> levels as mg/l in aquaculture literature preclude an unambiguous comparison 230 between data from these two fields. 231

232 Finally, understanding and reporting the provenance of the study species/population will be important to enable a more in depth assessment of CO<sub>2</sub> tolerance, i.e. whether animals 233 are wild caught, laboratory-bred or reared within an aquaculture setting (potentially already at 234 235 very high CO<sub>2</sub> when considered in a climate change experimental context). It is fair to say that many (though not all) laboratory-based climate change studies benefit from easy access to 236 study species available from aquaculture. The systematic selection of traits of interest by the 237 aquaculture industry, such as fast growth and resistance to pathogens, have inherently selected 238 for good performance under intensive farming conditions. In that context it is possible, and 239

even likely, that additional non-target traits have also been selected, potentially including those 240 involved in CO<sub>2</sub> tolerance. Indeed, enhanced CO<sub>2</sub> tolerance has been demonstrated in 241 selectively bred populations of the Sydney rock oyster, compared to its wild type congeners 242 (Parker et al., 2011, Parker et al., 2015, Thompson et al., 2015). Furthermore, as demonstrated 243 by Malvezzi et al. (2015) early life survival at elevated CO<sub>2</sub> concentrations can have a 244 significant additive genetic element (i.e. highly heritable), which under sufficient selection 245 246 pressure could elicit a strong and rapid evolutionary response. It is highly likely therefore, that aquaculture practices operating at elevated CO<sub>2</sub> concentrations would elicit sufficient selection 247 248 pressure to directly select for CO<sub>2</sub> tolerance during early life stages, leading to the rapid evolution of the population in just a few generations. Thus exploring the traits selected for in 249 broodstock within intensive aquaculture offers a fascinating opportunity to investigate multi-250 251 generational adaptation to CO<sub>2</sub> levels experienced under intensive production conditions in 252 aquaculture species. In addition, it will be vital to undertake multigenerational studies in order to discern the transgenerational acclimation to elevated CO<sub>2</sub> of different fish species with 253 respect to different behavioural (e.g. Welch et al., 2014) and physiological (e.g. Miller et al., 254 2012) endpoints. Combining the understanding from these two fields will therefore help 255 determine the physiological basis for CO<sub>2</sub> tolerance, determine its true ecological consequence 256 and determine its ecological impacts over relevant timescales. 257

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## 259 Conclusions

The yield from wild-capture fisheries has plateaued since the late 1980s and human consumption from aquaculture exceeded that from wild sources for the first time in 2014 (FAO, 2014). Furthermore, as stated previously, aquaculture is likely to be the only pathway for increasing seafood production in the future. Moving from a capture to a culture mentality 264 requires a shift in attitude that will require time, a luxury that is ill-afforded in the rapidly changing environment of the Anthropocene. Creating opportunities for the aquatic acidification 265 community and the aquaculture industry to work together should help to speed up this process 266 267 and enable the aquaculture industry to rapidly adapt by using better-informed decisions to; a) optimise the water chemistry conditions within intensive aquaculture to suit the species, and/or 268 b) select traits within the species to suit intensive aquaculture conditions. This will help address 269 the environmental, economic and social impacts of this developing sector towards a sustainable 270 intensification of production, enhancing food security and its resilience to climate change. 271 272 Equally, this cross-discipline interaction should also improve our capability to predict and mitigate the consequences of the changing chemistry for natural ecosystems in a future "high" 273 CO<sub>2</sub> world. 274

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Figure 1: Diagrammatic representation of the levels at which elevated carbon dioxide is 498 considered problematic within recirculating aquaculture systems (RAS) (caused by 499 accumulation of excreted CO<sub>2</sub> due to high stocking densities) and under global aquatic 500 501 acidification (marine and freshwater; caused by rising atmospheric CO<sub>2</sub>). Numbered arrows, and corresponding key, indicate the levels at which CO<sub>2</sub> is demonstrated to have significant 502 impacts on fish development, physiology and behaviour. The expanded view on the right side 503 504 highlights CO<sub>2</sub> levels in relation to climate change scenarios in greater detail (0–3,000 µatm or 0–4 mg/l). Conversion of  $CO_2$  levels between µatm and mg/l in this diagram is based on 35 505 506 psu seawater at 15°C. Fish images Kovalevska & Kazakov maksim /shutterstock.com. 507 References corresponding to numbered arrows indicating levels of CO<sub>2</sub> shown to have a significant impact of fish development, physiology or behaviour; 1) Hamilton et al. (2014), 508 509 Jutfelt and Hedgärde (2013), Simpson et al. (2011), Nilsson et al. (2012); 2) Chambers et al. 510 (2014), Frommel et al. (2012), Frommel et al. (2014), Maneja et al. (2014), Tseng et al. (2013); 3) Esbaugh et al. (2016), Esbaugh et al. (2012), Heuer et al. (2012); 4) Pope et al. 511 (2014); 5) Ou et al. (2015); 6) Michaelidis et al. (2007); 7) Tirsgaard et al. (2015); & 8) 512

513 Seidelin *et al.* (2001).

Figure 2: Schematic representation of the conversion of 1 mg L<sup>-1</sup> dissolved CO<sub>2</sub> 514 concentration into partial pressure (µatm) at a range of different temperatures and salinities. 515 This shows the very large influence of temperature in particular (up to 3.2-fold higher partial 516 517 pressure at the warmest temperature compared to the coolest) but also salinity (up to 26% higher partial pressure at the highest salinity compared to freshwater) on the CO<sub>2</sub> partial 518 pressure due to the impact these abiotic factors have on the solubility of CO<sub>2</sub> in water (Weiss, 519 1974, Dickson, 2011). Conversion of dissolved CO<sub>2</sub> in mg  $L^{-1}$  to partial pressure in µatm 520 were undertaken using the CO2SYS program (Pierrot et al., 2006), using dissociation 521 522 constants from Mehrbach et al. (1973), refit by Dickson and Millero (1987), and KSO4 using Dickson (1990), with values for CO<sub>2</sub> solubility at different temperatures and salinities 523 checked against Weiss (1974). 524

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## 533 Author contributions

- R.W.W. won the funding for aquaculture and aquatic acidification projects that stimulated
- this article and produced figure 2. R.E led the formulation of the paper and produced figure 1.
- 536 M.U compiled the initial draft. All authors contributed equally to discussions, figure
- 537 development, editing and production of the final manuscript.

# 538 **Competing financial interests**

539 The authors declare no competing financial interests.



